

Exploring Big Data for a Deeper Understanding of Electrocatalyst Behavior

- starting...

ElectroCat Laboratory



NATIONAL INSTITUTE OF CHEMISTRY

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Laboratory for ElectroCatalysis

Established in 2020:

Catalyst Synthesis and Characterization

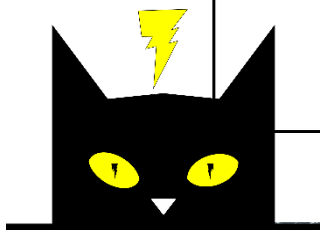
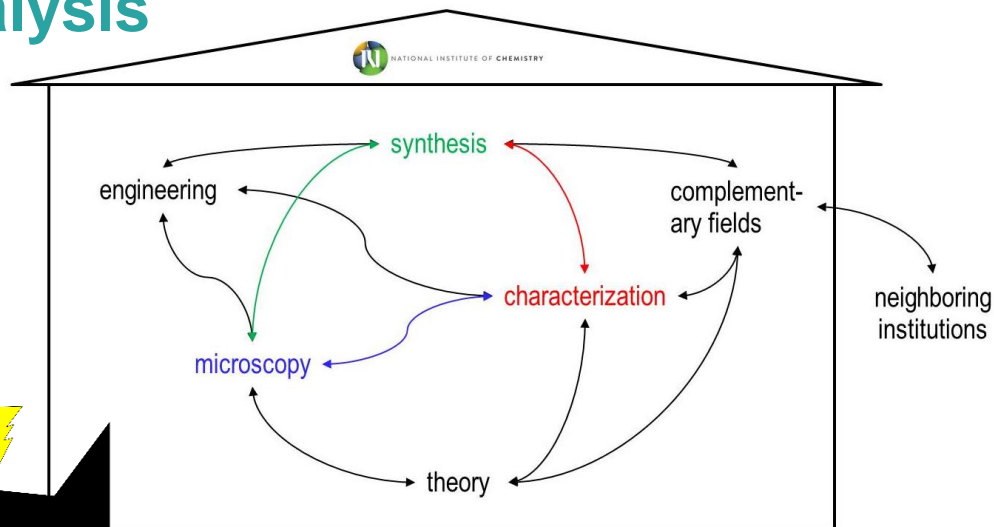
Advanced electrochemical characterization

Electron microscopy

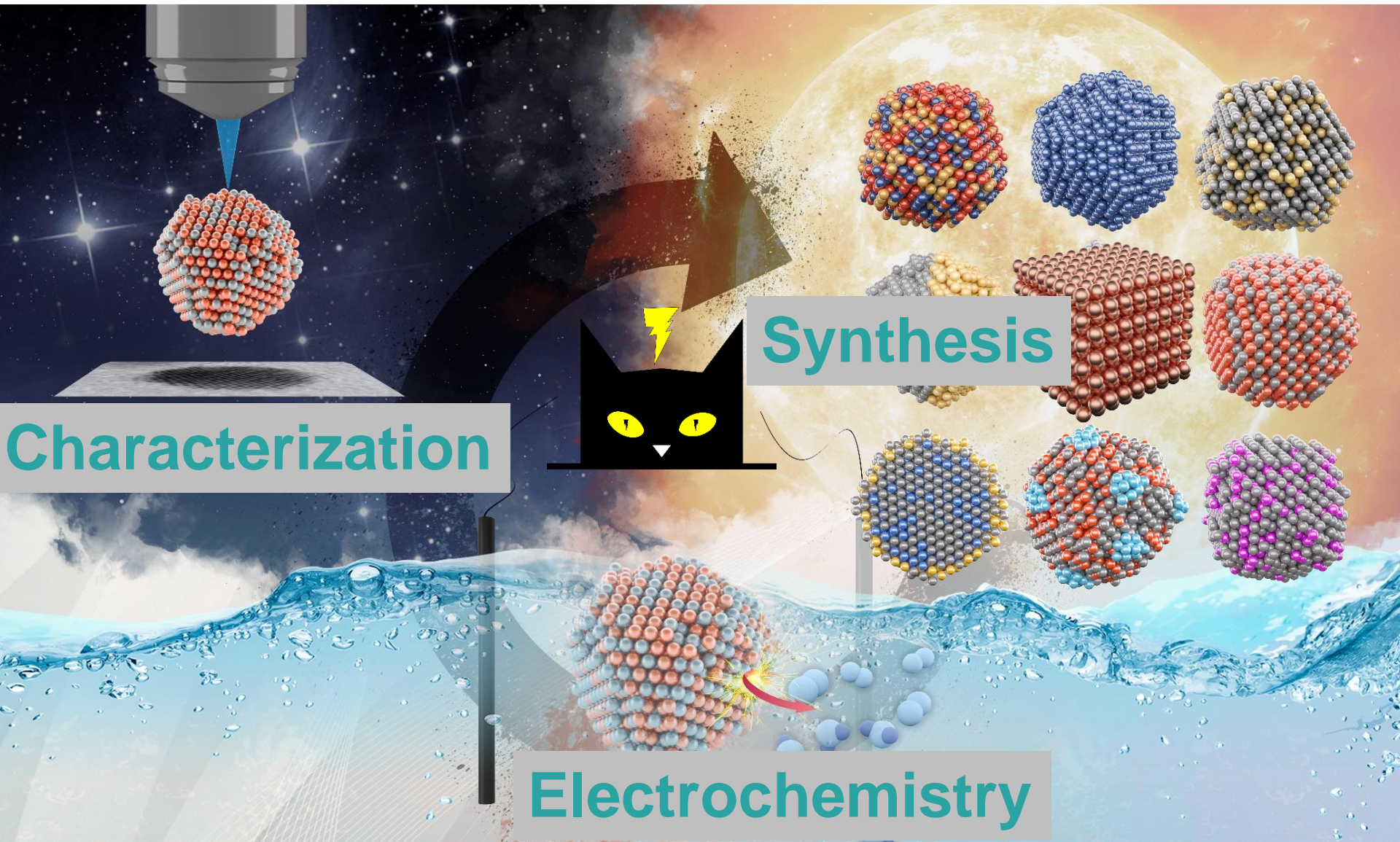
Recycling

Electrosynthesis

Collaborations...



We support each other to be the best versions of ourselves.



Characterization

Synthesis

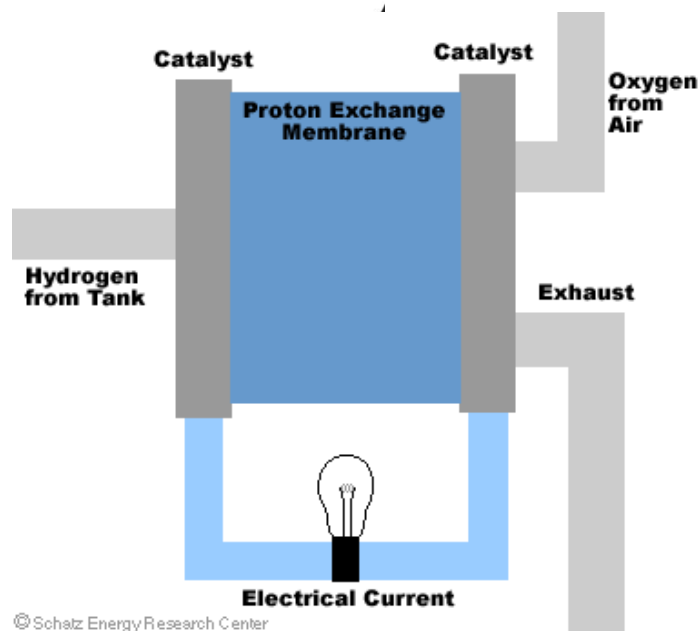
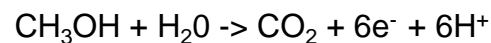
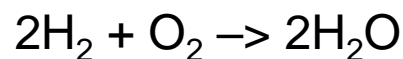
Electrochemistry

Electrocatalysis

Very intriguing chemistry that is happening at the metal-electrolyte interface!

The electrons are decoupled from the reactants by splitting the reaction in two.

Main difference to **catalysis** is the double layer! *Old field...reinvigorated!*

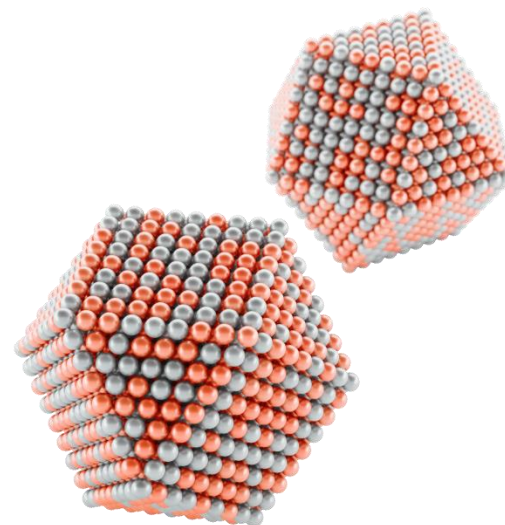


It has relevance to the future possibilities of the energy infrastructure!

What are nanoparticles?

- Nanoparticles are particles ranging in size from 1 to 100 nanometers.
- Hair thickness is approximately one hundred thousand (100 000) nanometers.
- At these sizes, the properties change (large area and more...).
- Nanoparticles are composed of relatively few atoms
 - 2 nm -> 3000 to 5000, 4 nm -> 20000 atoms
 - Nanozymes
 - Electro-Catalysts

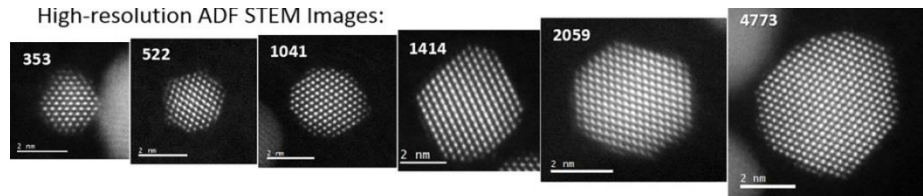
Compared to molecule catalysts (homogeneous) or viruses that have atomically defined structures ours are very complex!



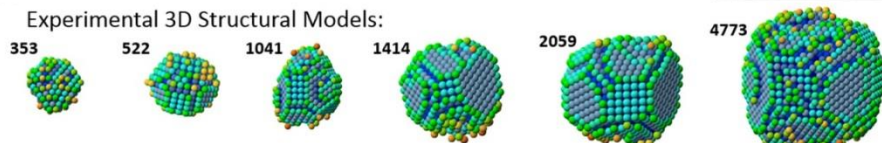
Difference between homo- and heterogeneous-catalysis

No nanoparticle is perfect!

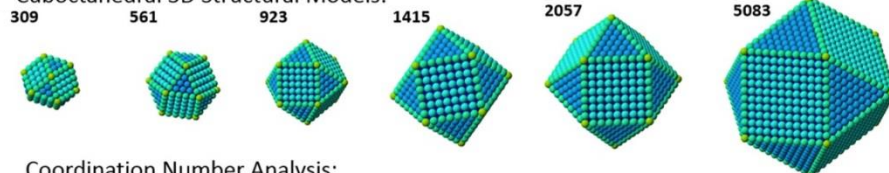
High-resolution ADF STEM Images:



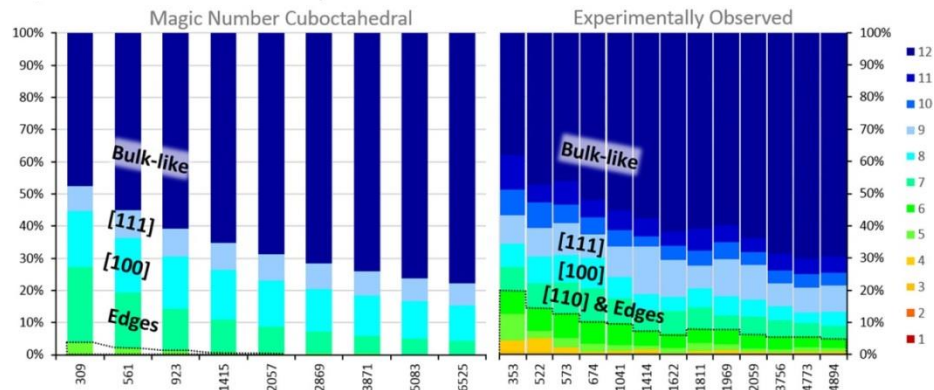
Experimental 3D Structural Models:



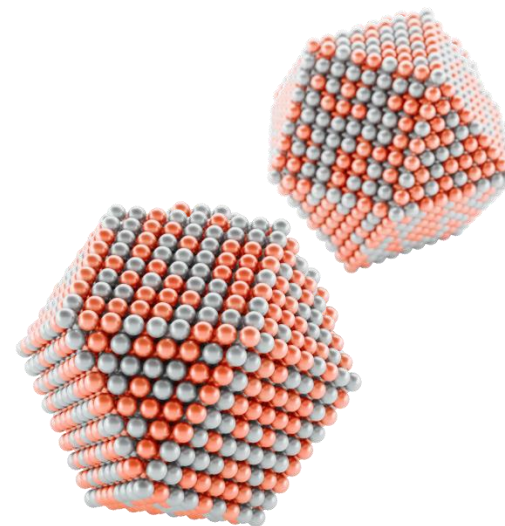
Cuboctahedral 3D Structural Models:



Coordination Number Analysis:

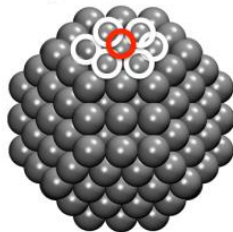


No two particles that we will examine will be alike!



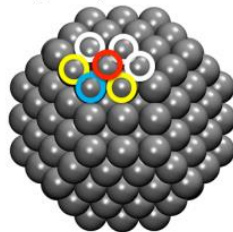
Nanostructured Pt-alloys ORR electrocatalysts

a) 111TC (cn = 9)



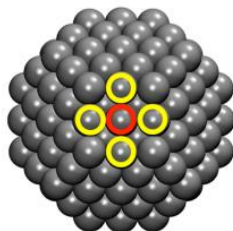
$$\overline{CN} = \frac{9 \times 6 + 12 \times 3}{12} = 7.50$$

b) 111TM (cn = 9)



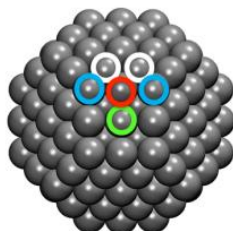
$$\overline{CN} = \frac{6 \times 1 + 7 \times 2 + 9 \times 3 + 12 \times 3}{12} = 6.92$$

c) 100TC (cn = 8)



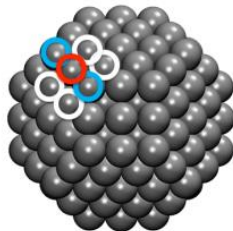
$$\overline{CN} = \frac{7 \times 4 + 12 \times 4}{12} = 6.33$$

d) 100E (cn = 7)



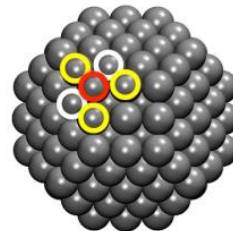
$$\overline{CN} = \frac{6 \times 2 + 8 \times 1 + 9 \times 2 + 12 \times 2}{12} = 5.17$$

e) 111E (cn = 7)



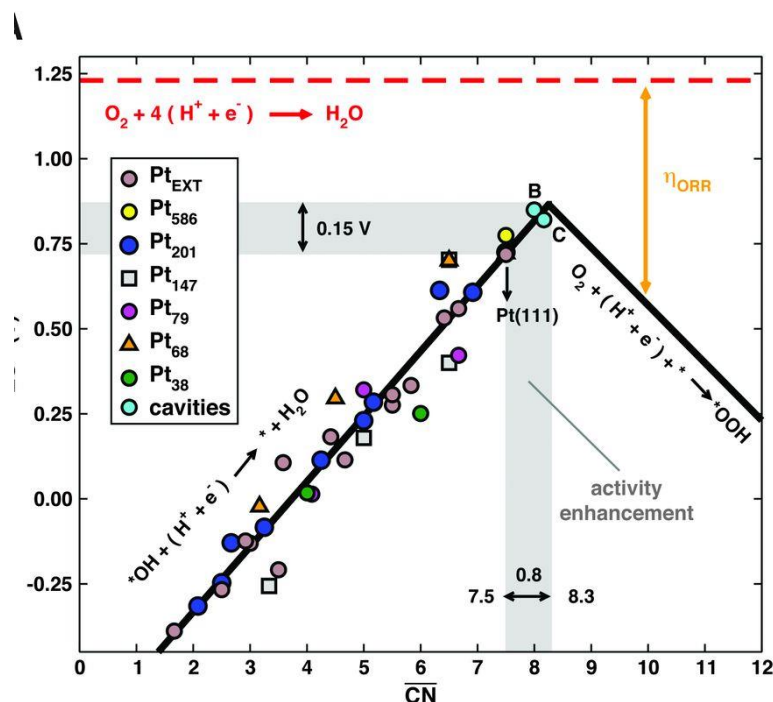
$$\overline{CN} = \frac{6 \times 2 + 9 \times 4 + 12 \times 1}{12} = 5.00$$

f) corner (cn = 6)

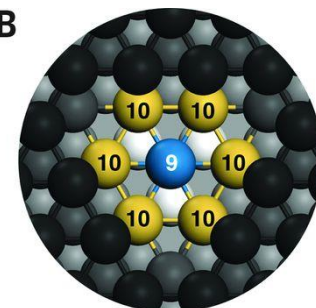


$$\overline{CN} = \frac{7 \times 3 + 9 \times 2 + 12 \times 1}{12} = 4.25$$

ad

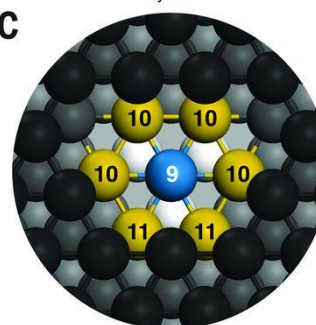


B



$$\overline{CN}_{\text{cavity-B}} = 8.00$$

C

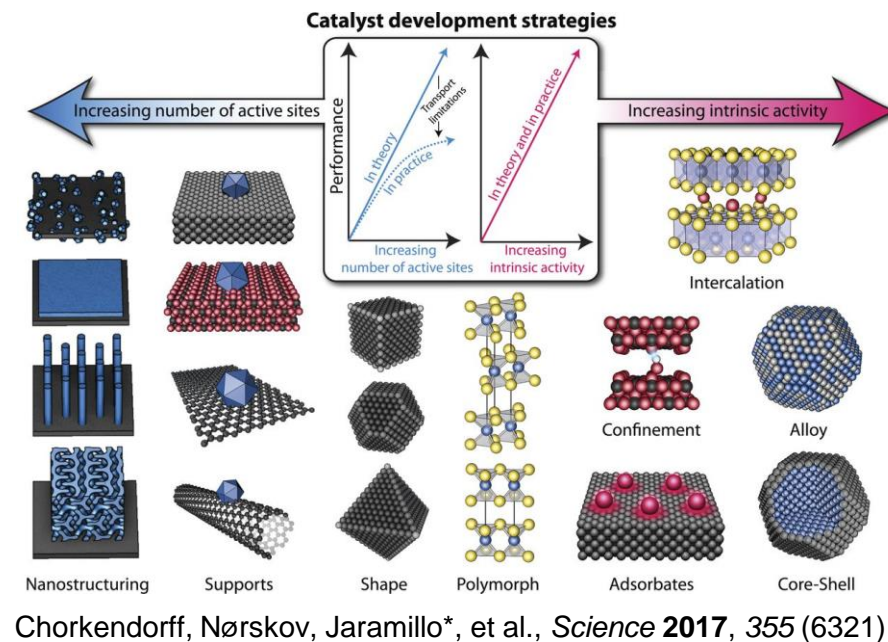


$$\overline{CN}_{\text{cavity-C}} = 8.17$$

What defines metal electrocatalyst performance?

- double layer structure (what defines that besides the electrolyte...)

- nature of the metal
- size
- morphology (surface facets)
- structure
- composition (presence of second metal)
- presence of defects, steps, kinks, ad-atoms
- alloying ligand and/or strain
- support
- confinement, proximity, ensemble
- surface patterning, nature of electrolyte
- etc.

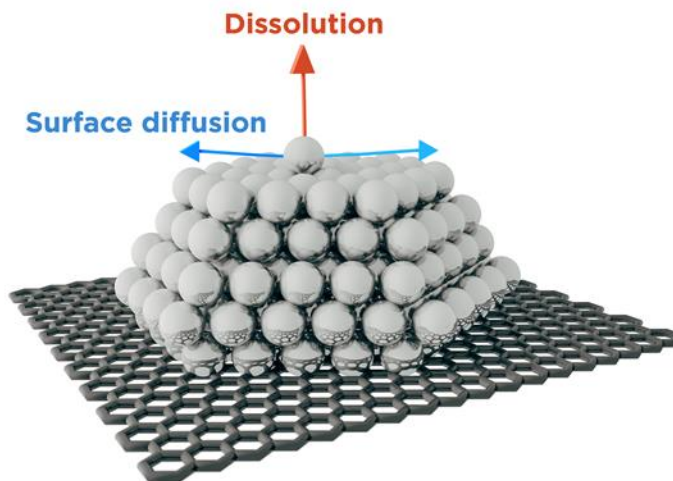


Practically everything we can think of!

Nobel laureate Wolfgang Pauli said, "God made the bulk; surfaces were invented by the devil".

The devil is in the details.

Structure-activity relationship on atomic level is a **state-of-the-art** in electrocatalysis



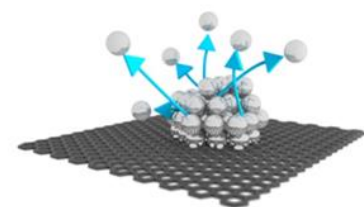
However we can not synthesize identical (perfect) nanoparticles as a model systems!

Structure-stability relationship at the atomic level is much less explored!

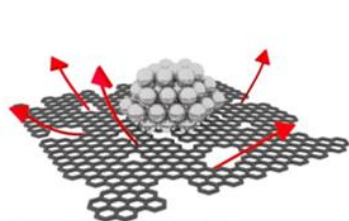
And we have great methods that have (close to) atomic resolution.

It is indeed much harder to measure activity one atom then see how it moves...

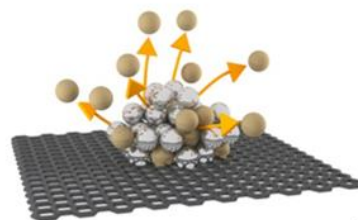
Electrochemical degradation mechanisms of nanoparticles



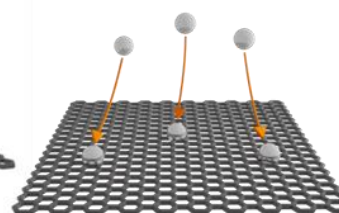
Platinum Dissolution



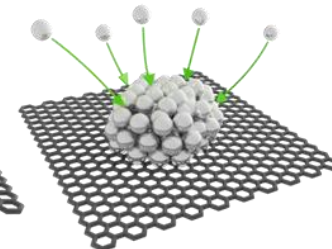
Carbon Corrosion



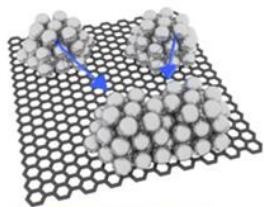
Dealloying



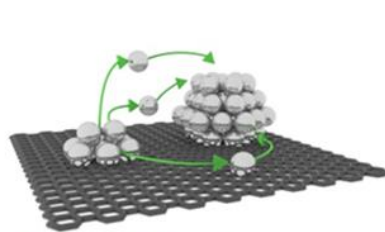
Nucleation



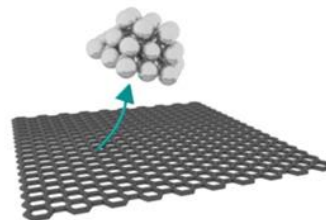
Growth



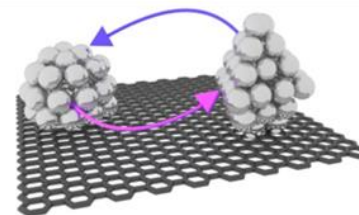
Agglomeration



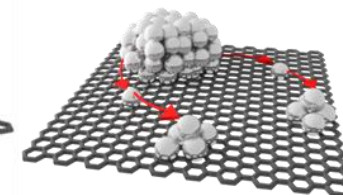
Ostwald Ripening



Particle Detachment



Reshaping

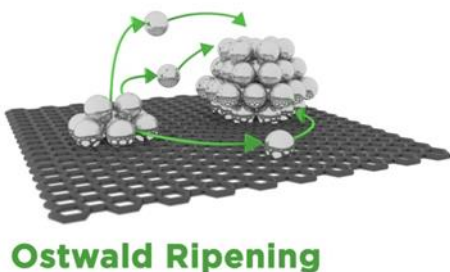


Dispersion

Complex interplay of different mechanisms resulting in the change of active surface area (j_a and ESA)!

Degradation of Pt-based fuel cell catalyst

Postmortem analysis?
No!

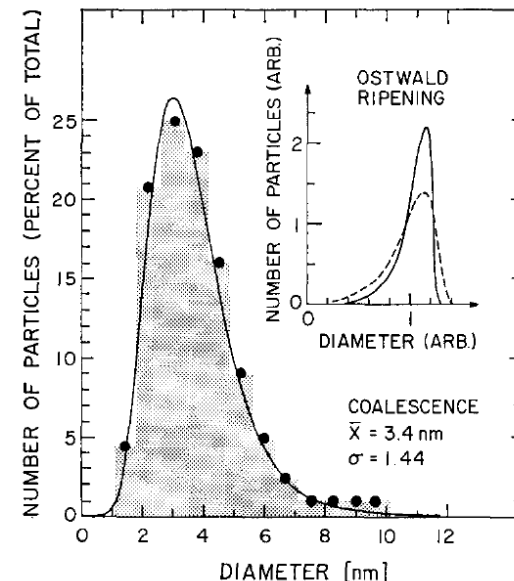
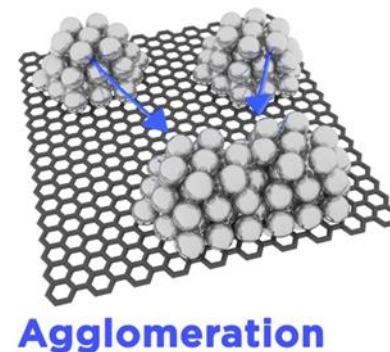


How to distinguish between them?

Bulk methods like particle size distribution (PSD) obtained by analyzed TEM images and XRD offer only averaged information:

It was shown that PSDs with **tails to larger particle sizes** are associated with particle growth via **migration and coalescence**.

In contrast, a PSD indicative of growth through electrochemical **Ostwald ripening** involves a peak toward large particle sizes with **tailing to smaller sizes**.



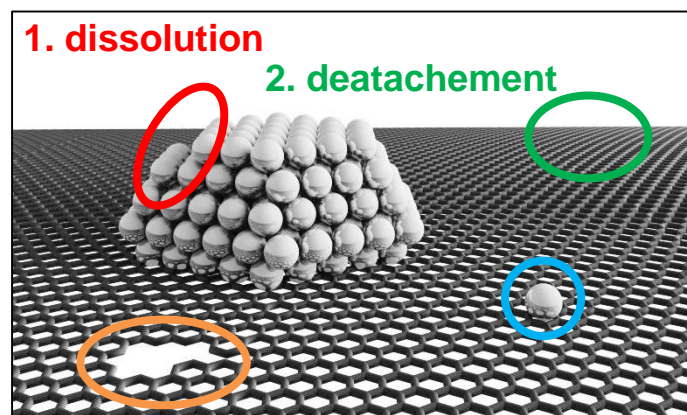
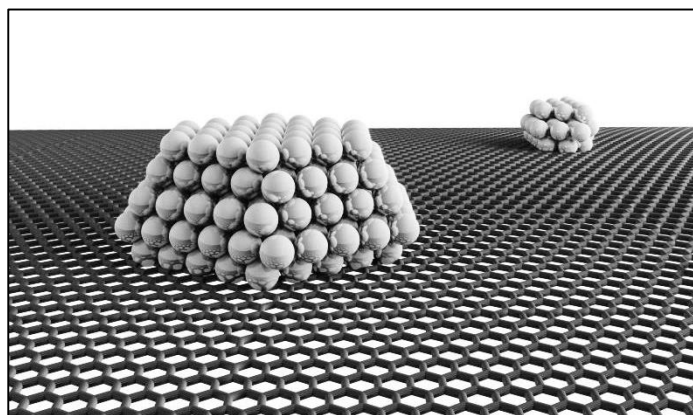
Problem: the simple models assume that only one growth mode occurs! – not true!

C.G. Granqvist, R.A. Buhrman, Journal of Catalysis , 1976, 42, 477-479.

Identical Location Electron Microscopy

Spot the difference at the atomic level

Can you spot all 4 differences and mark them?



3. carbon corrosion

4. redeposition

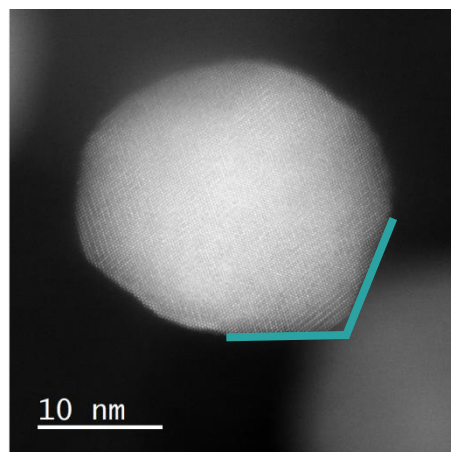
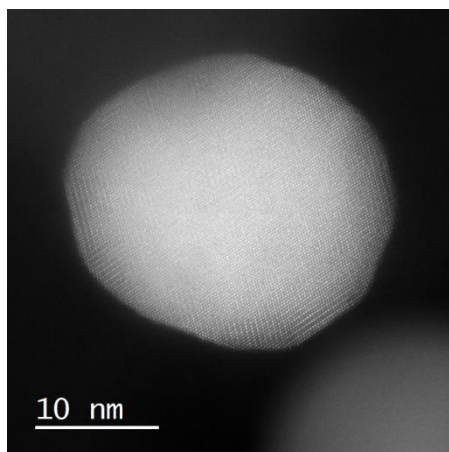
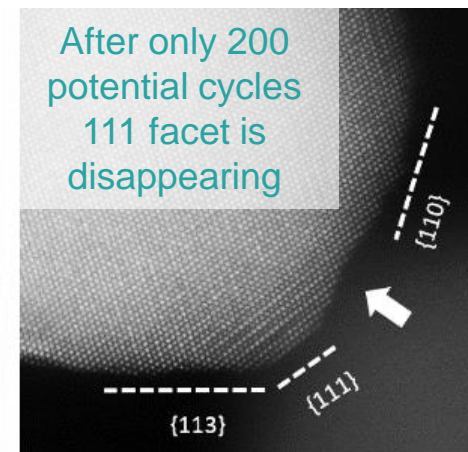
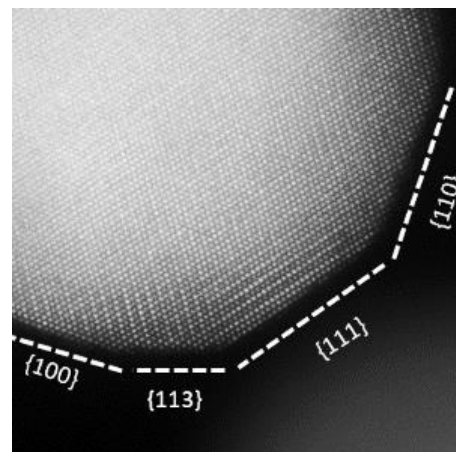
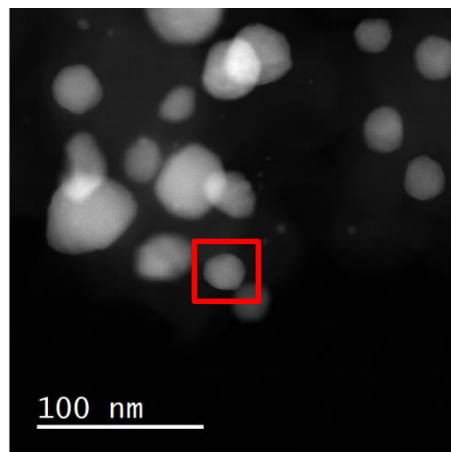
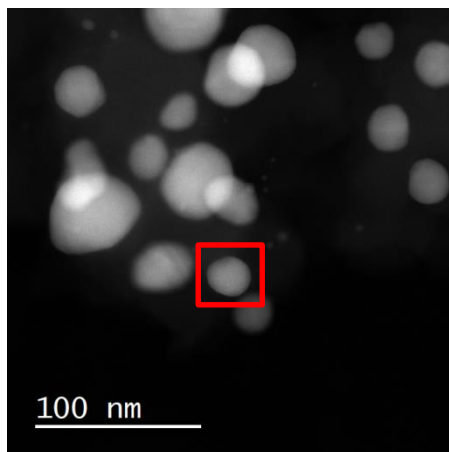
See the two (electrocatalysts) images before and after aging.

Because the history of the location's physical characteristics is known, reliable conclusions on very complex restructuring events such as degradation mechanisms can be drawn!

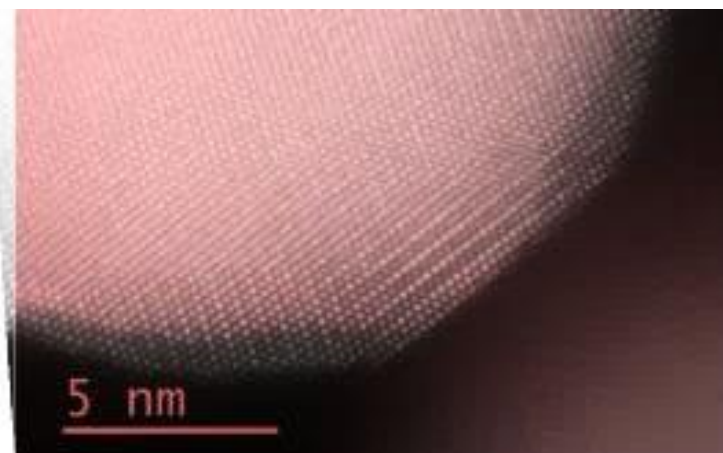
Identical Location ...



PtM nanoparticles reshaping



Overlay



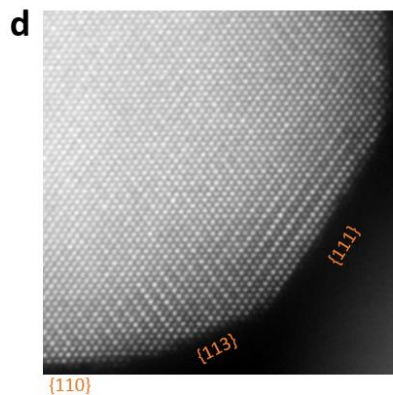
IL-TEM 0,1 M HClO₄
 200 cyc 0.05 – 1.2 V vs. RHE, 300 mV/s, Ar

Computer simulation

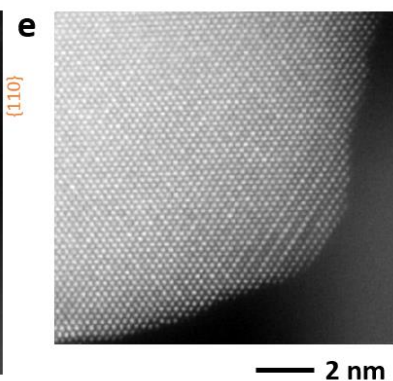
Atomic physical model - KMC!

Pavlisic & Hodnik et al. *ACS Catalysis* (2016), 6(8), 5530

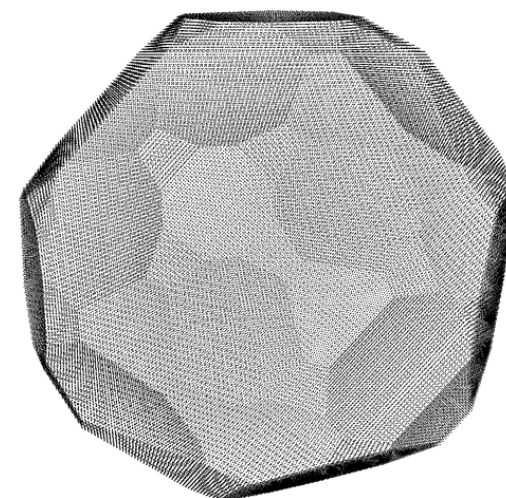
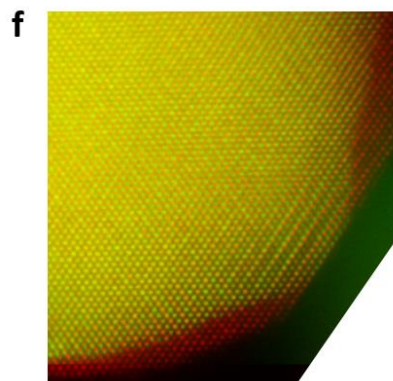
STEM before



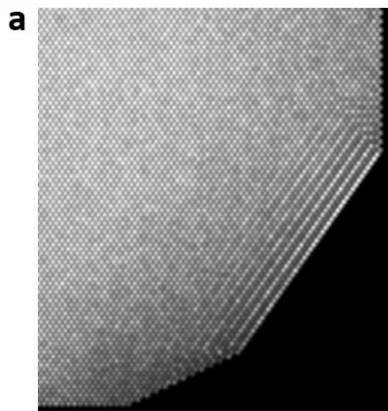
STEM after



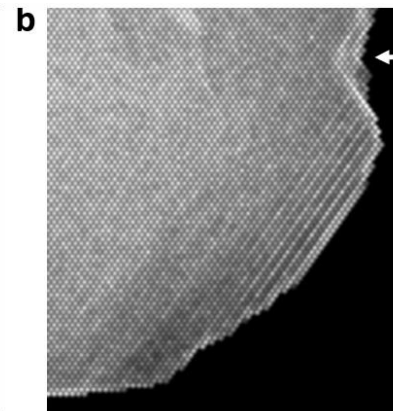
overlay



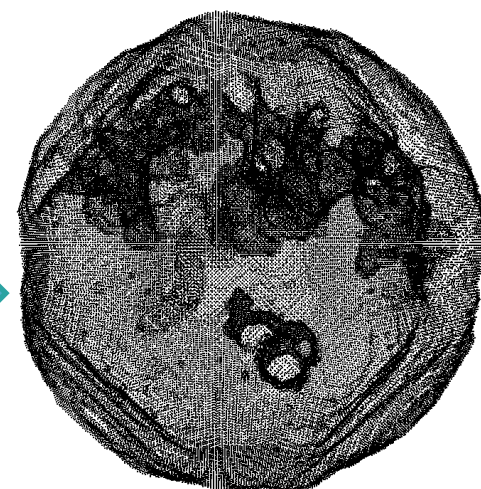
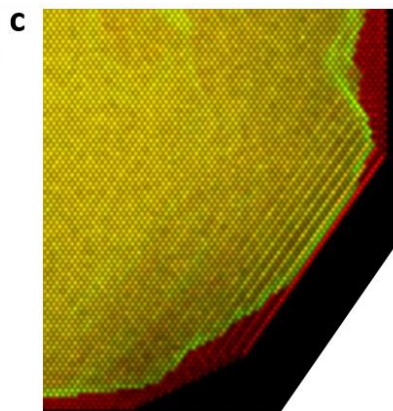
Simulation before



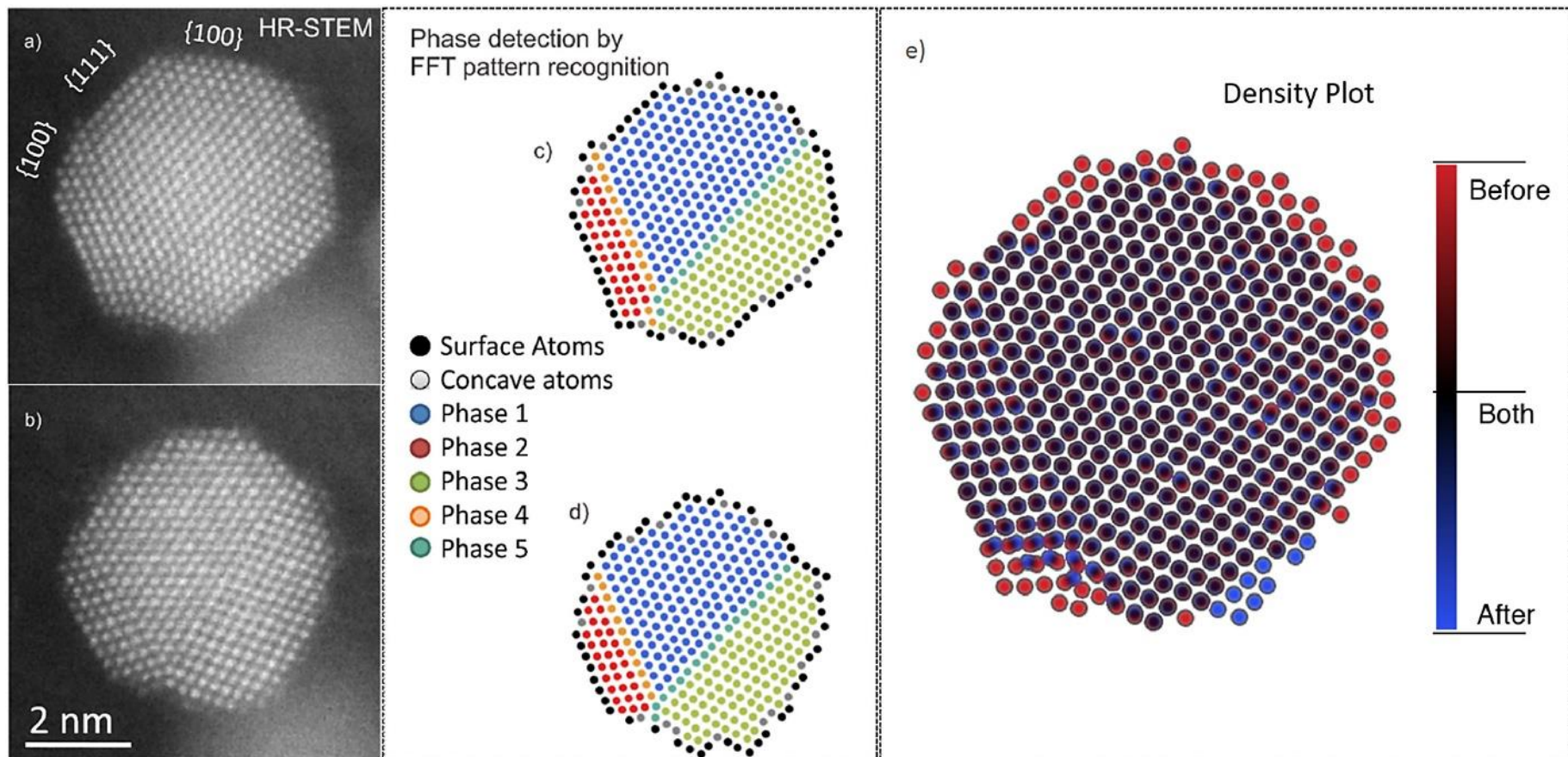
Simulation after



overlay



PtM nanoparticles reshaping



Problems – or my excuses

- How to measure reproducibly?
- How to trust somebody else's measurements if we even do not trust ours!
 - iR compensation, catalyst film quality, loading effect, contamination, sample preparation, treatment of data (data massage), use of different potentiostats, reference electrode, ...
- Electrochemical signal is an average of signals (bulk)
- Effects of mass transport or local environment like pH
- Impurities (electrochemistry or even some single atom in graphene in TEM)
- Stability issues (electrochemistry or el. beam induced)
- The problem of error and reproducibility (a lot of effects besides just operator; karma or we saw Thursdays and Decembers are the best!)

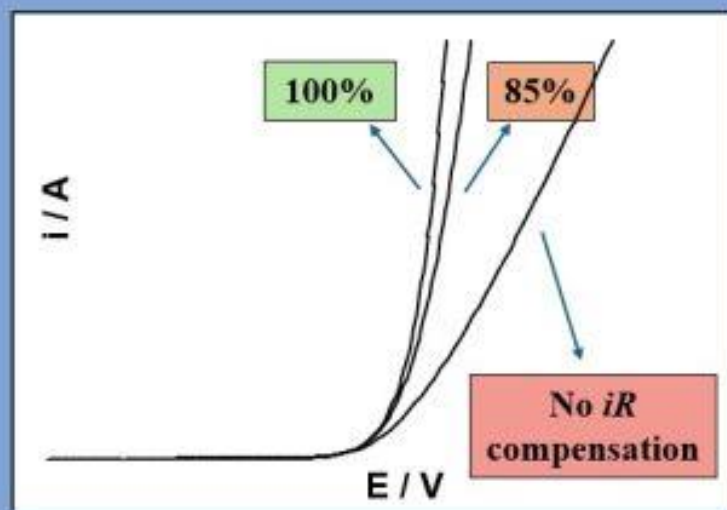
Problems – electrochemistry

Electrochemical tip #5

“

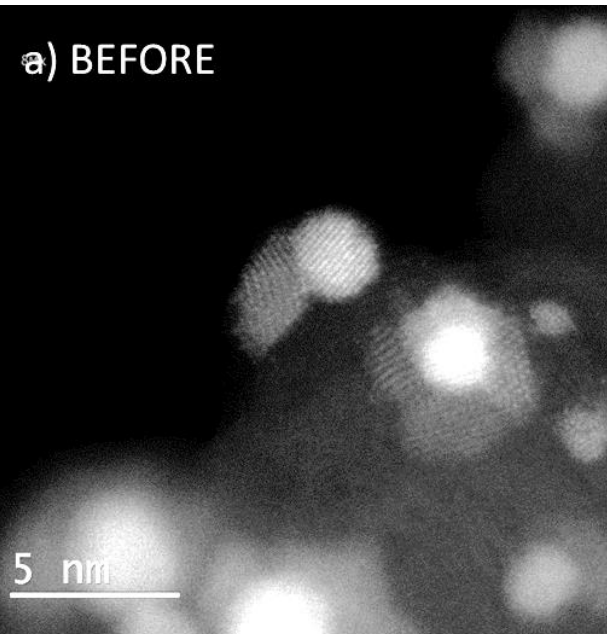
iR compensation is essential,
but it requires careful considerations

”

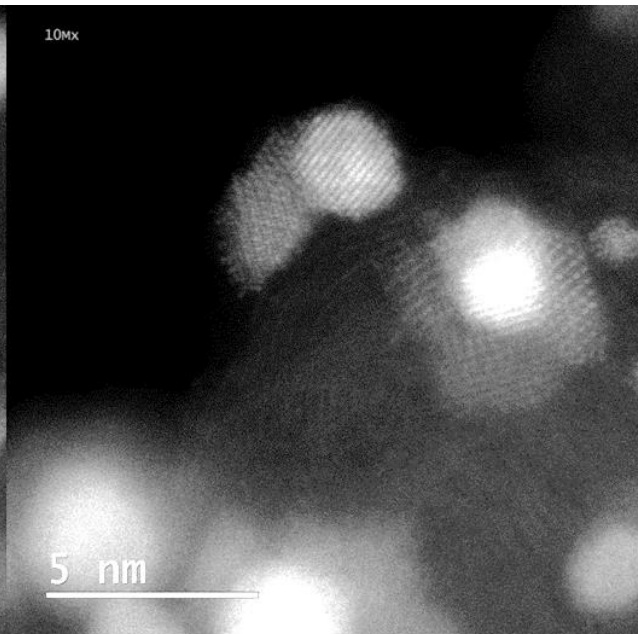


D. Galyamin

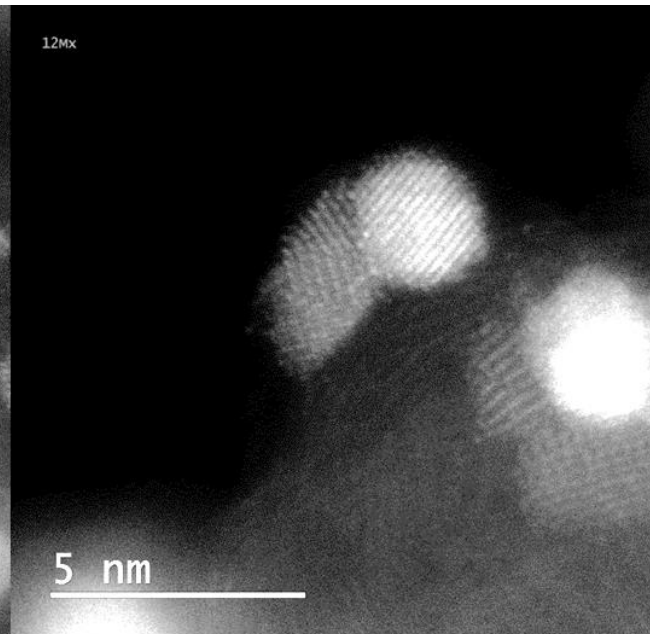
a) BEFORE



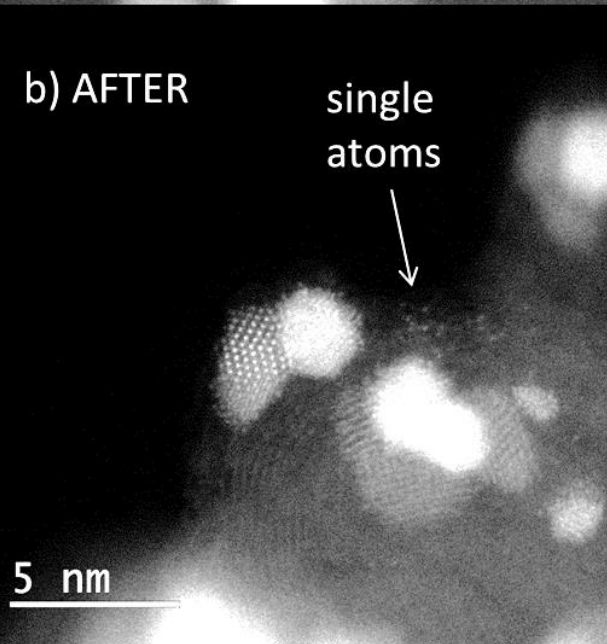
10Mx



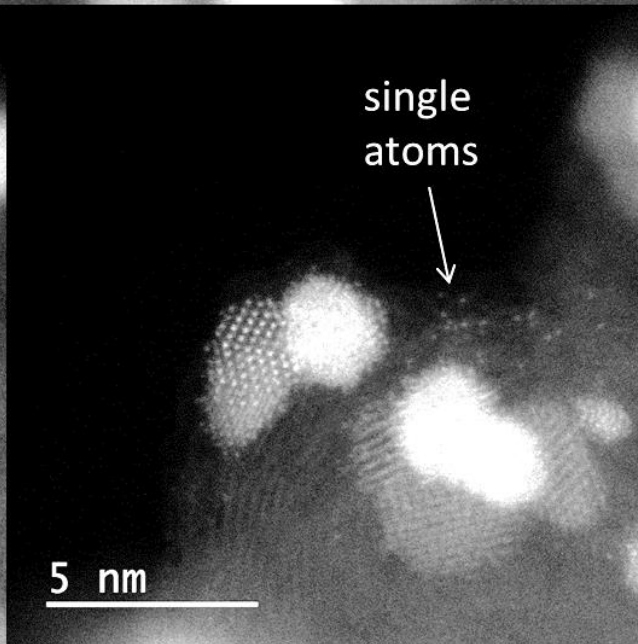
12Mx



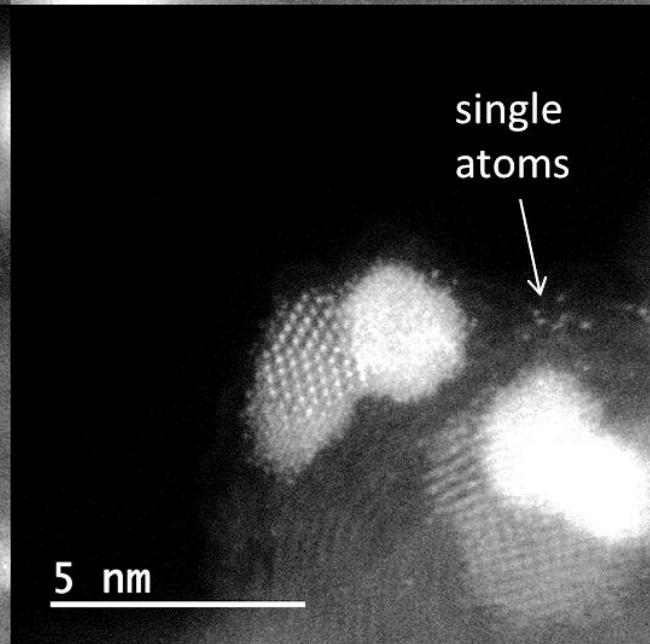
b) AFTER



single atoms



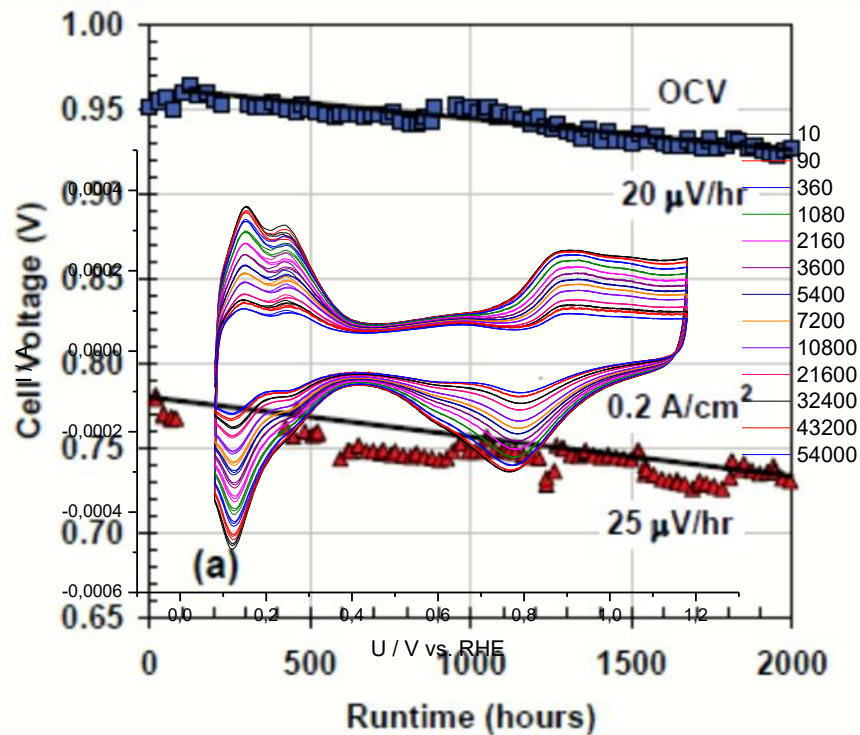
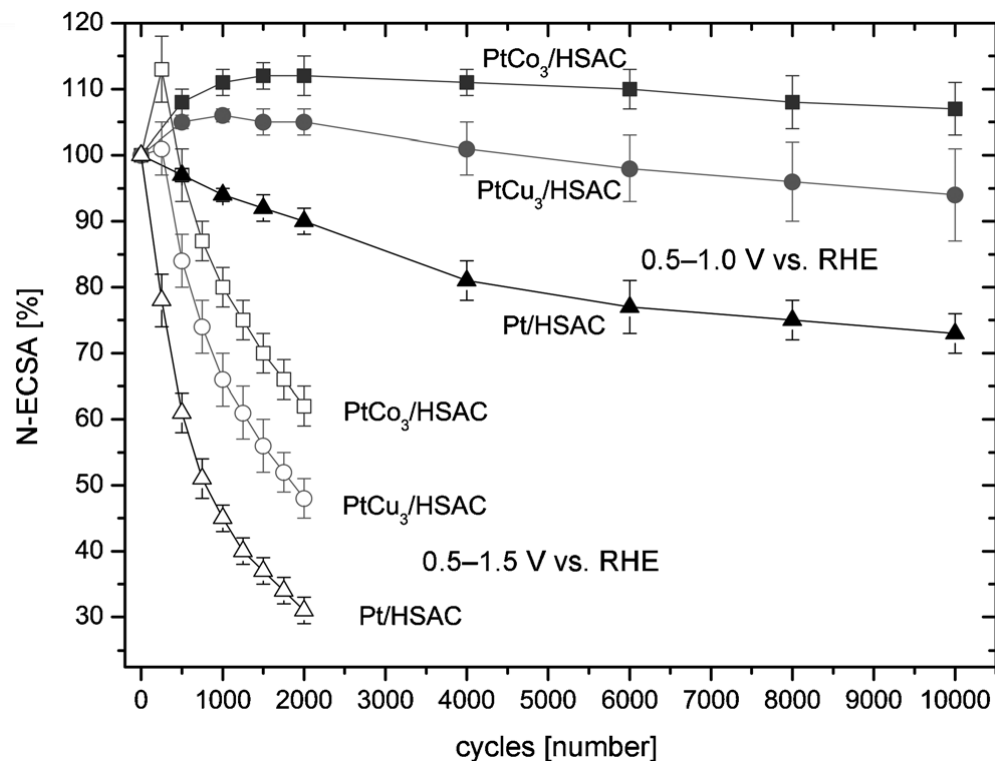
single atoms



single atoms

Proton exchange membrane fuel cell

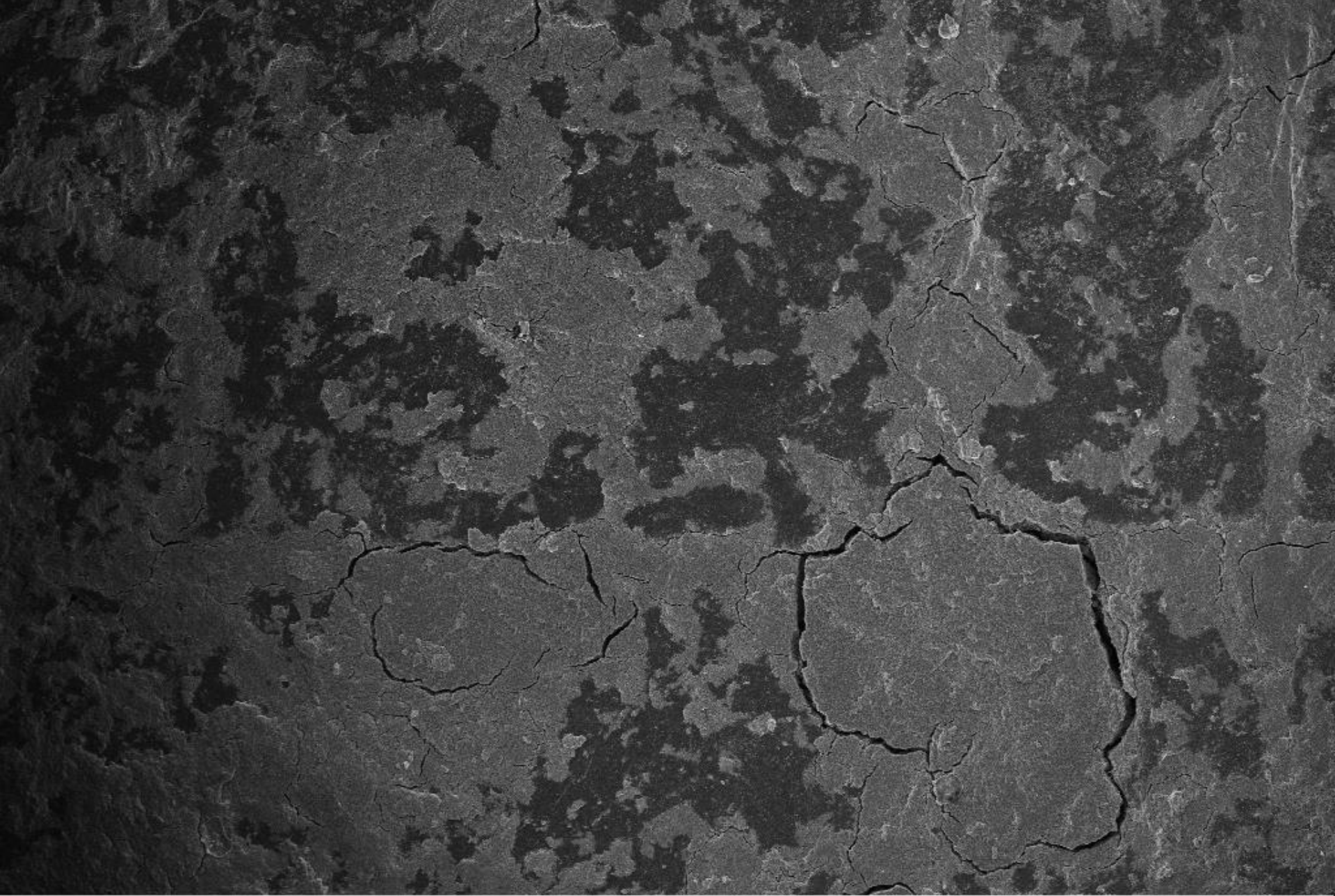
Device inherently has a averaged behavior



Frédéric Hasché, Activity, *ChemCatChem*, 3, (2011) 1805

Ferreira, *JES*, 2005, 152

Is it possible to avoid this – we must truly understand fundamentals first → ex-situ tests

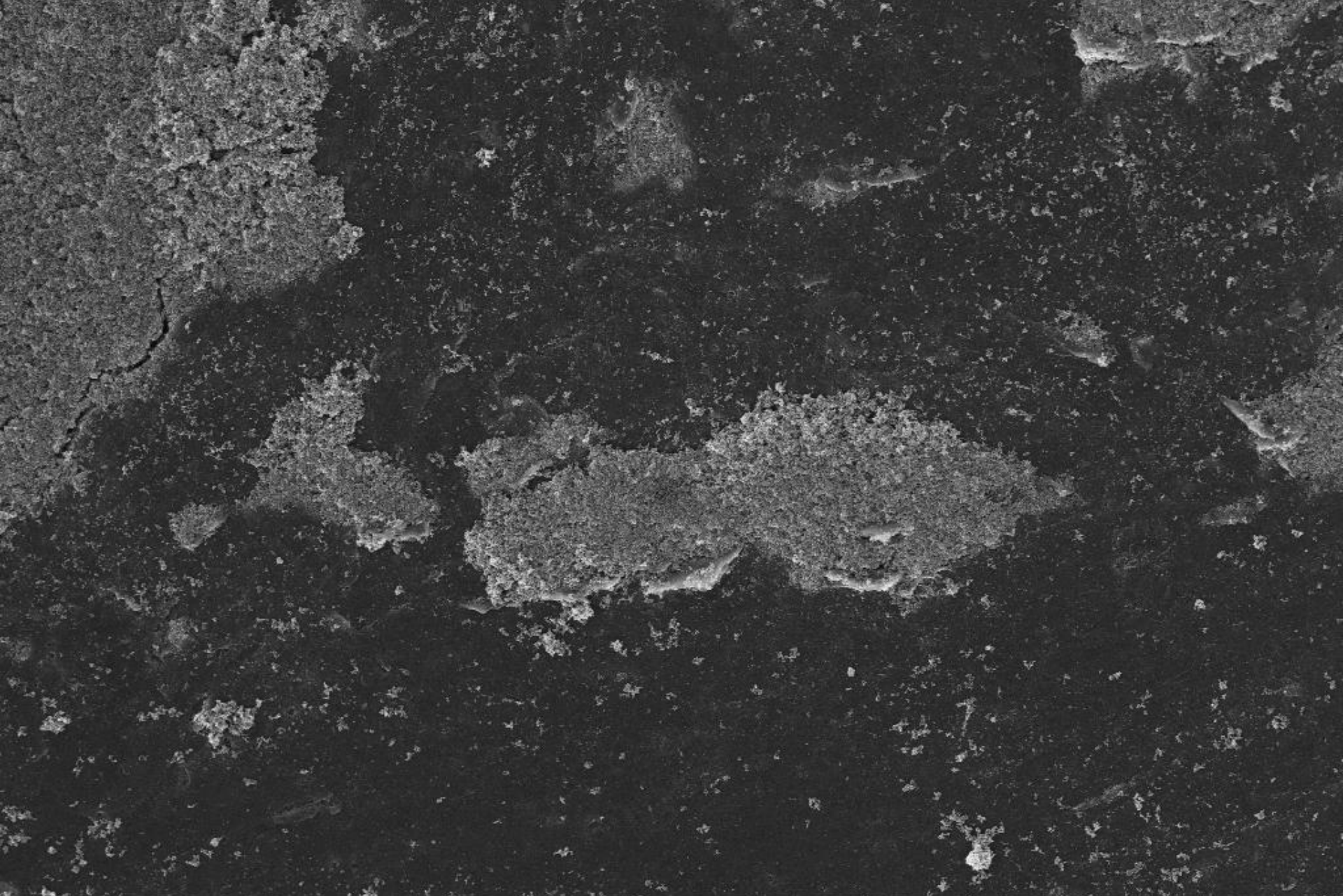


100 μm

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.87e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μm File Name = Denora_Pt_22.tif

Date : 23 Oct 2012



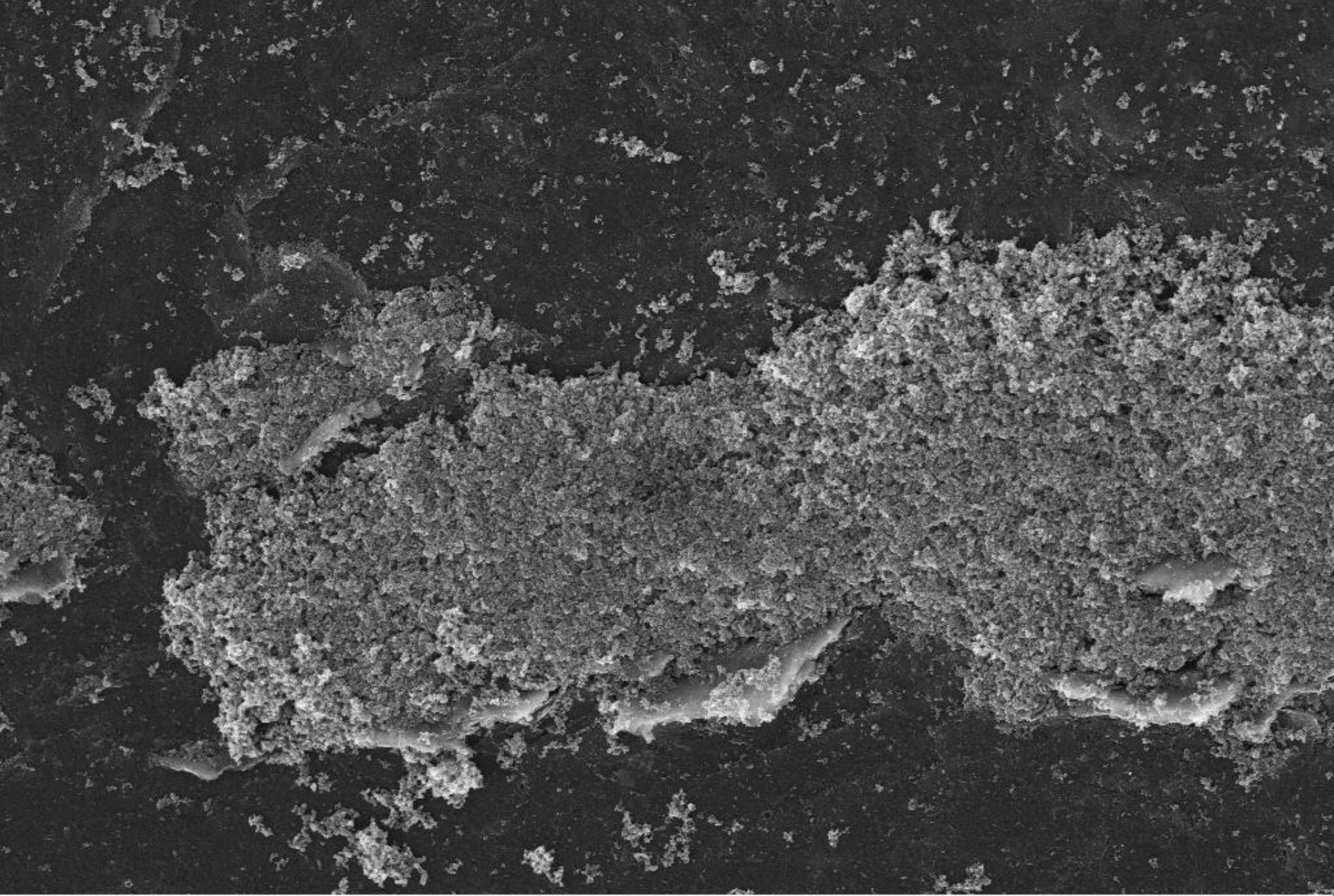


10 μm

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.81e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μm File Name = Denora_Pt_31.tif

Date : 23 Oct 2012



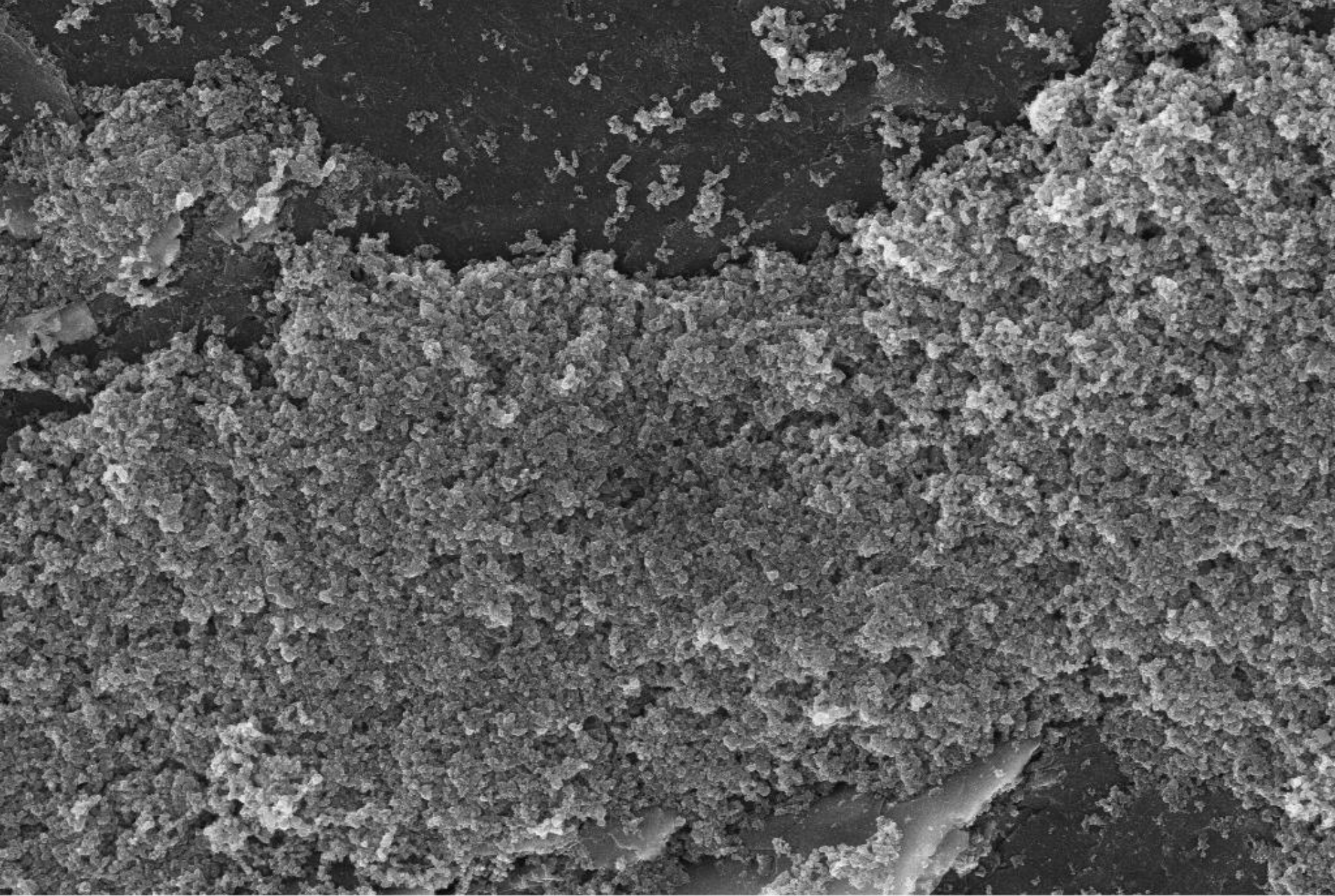


1 μm
H

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.85e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μm File Name = Denora_Pt_30.tif

Date : 23 Oct 2012



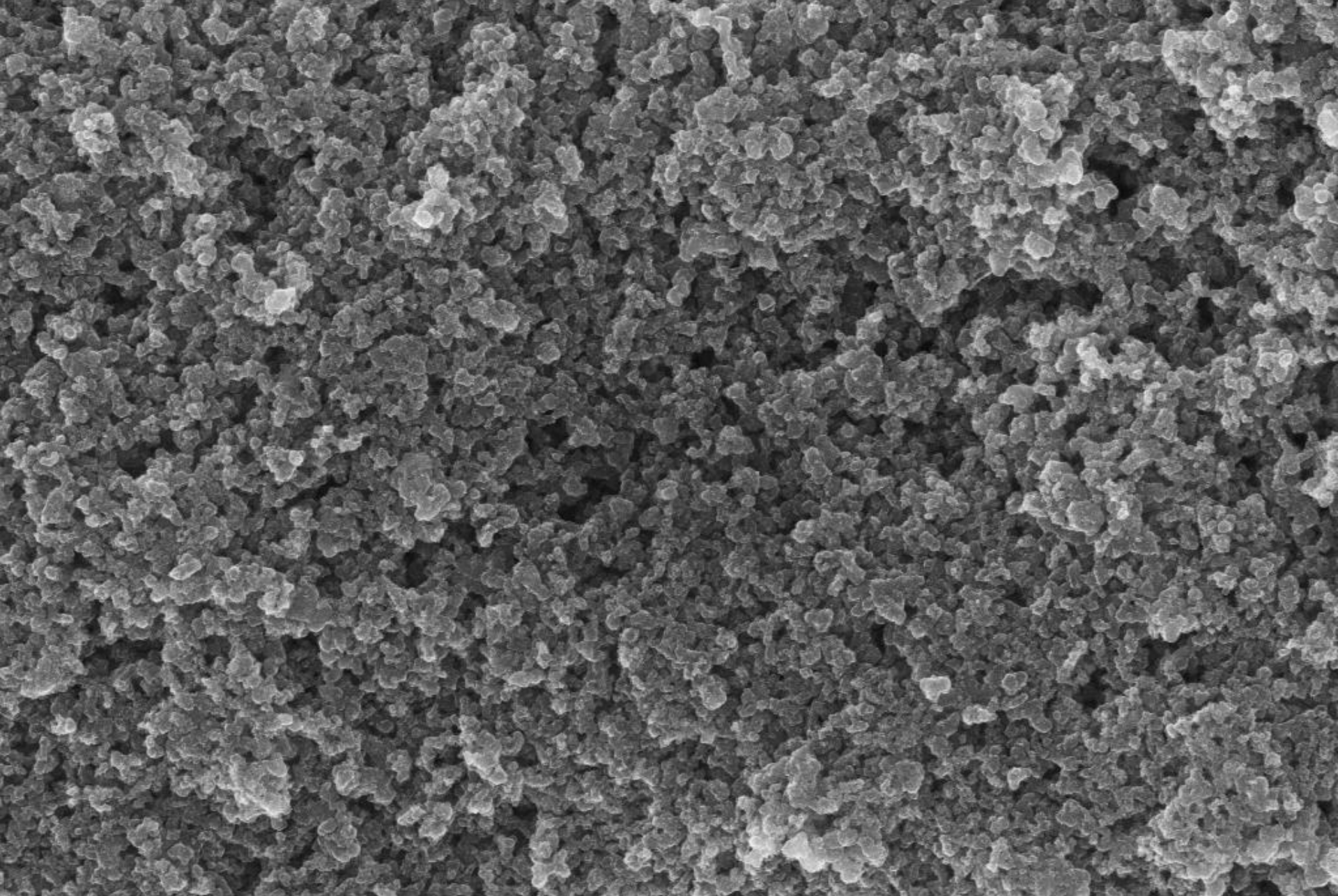


1 μm
|-----|

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.83e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μm File Name = Denora_Pt_29.tif

Date : 23 Oct 2012





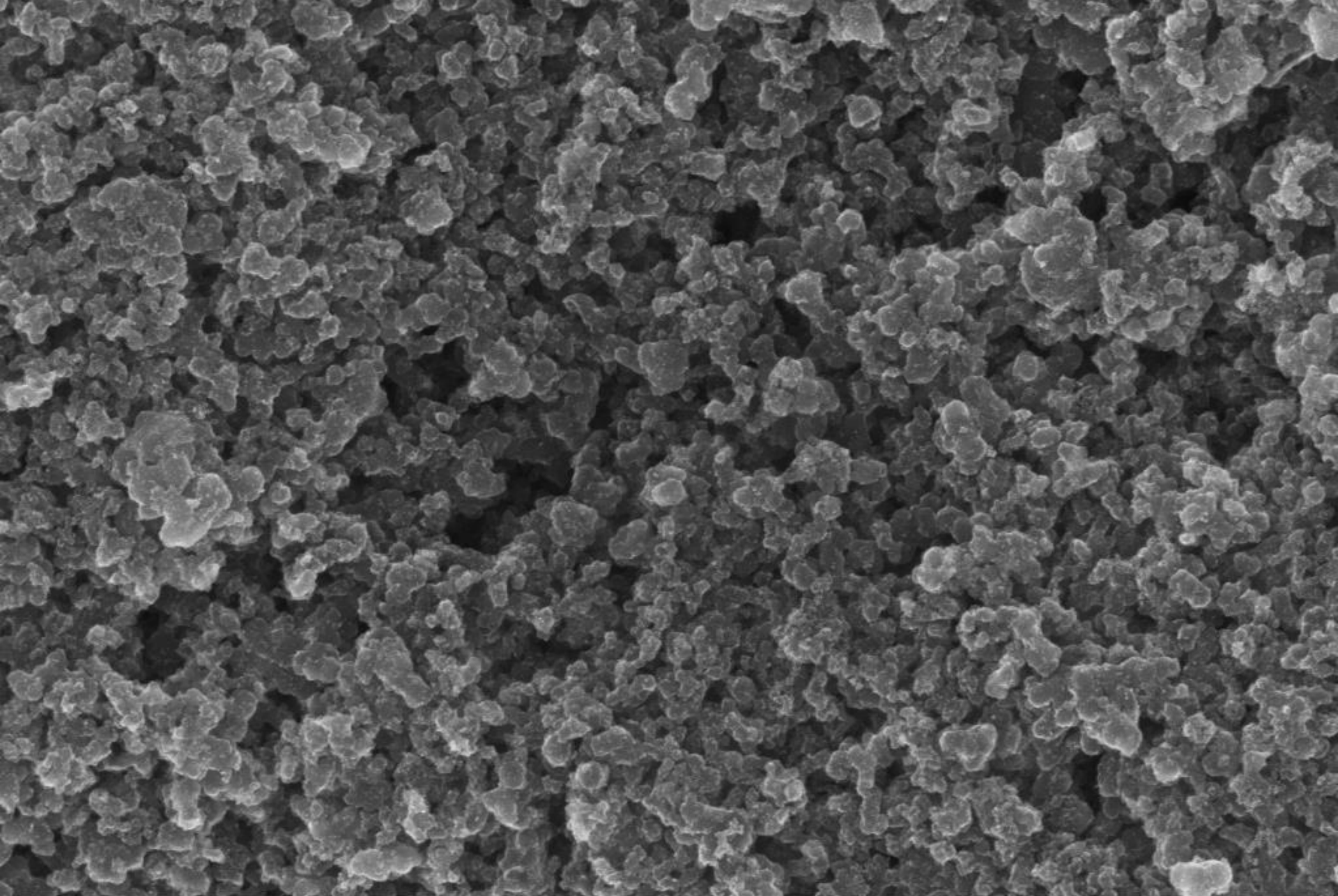
1 μm



EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.85e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μm File Name = Denora_Pt_28.tif

Date : 23 Oct 2012





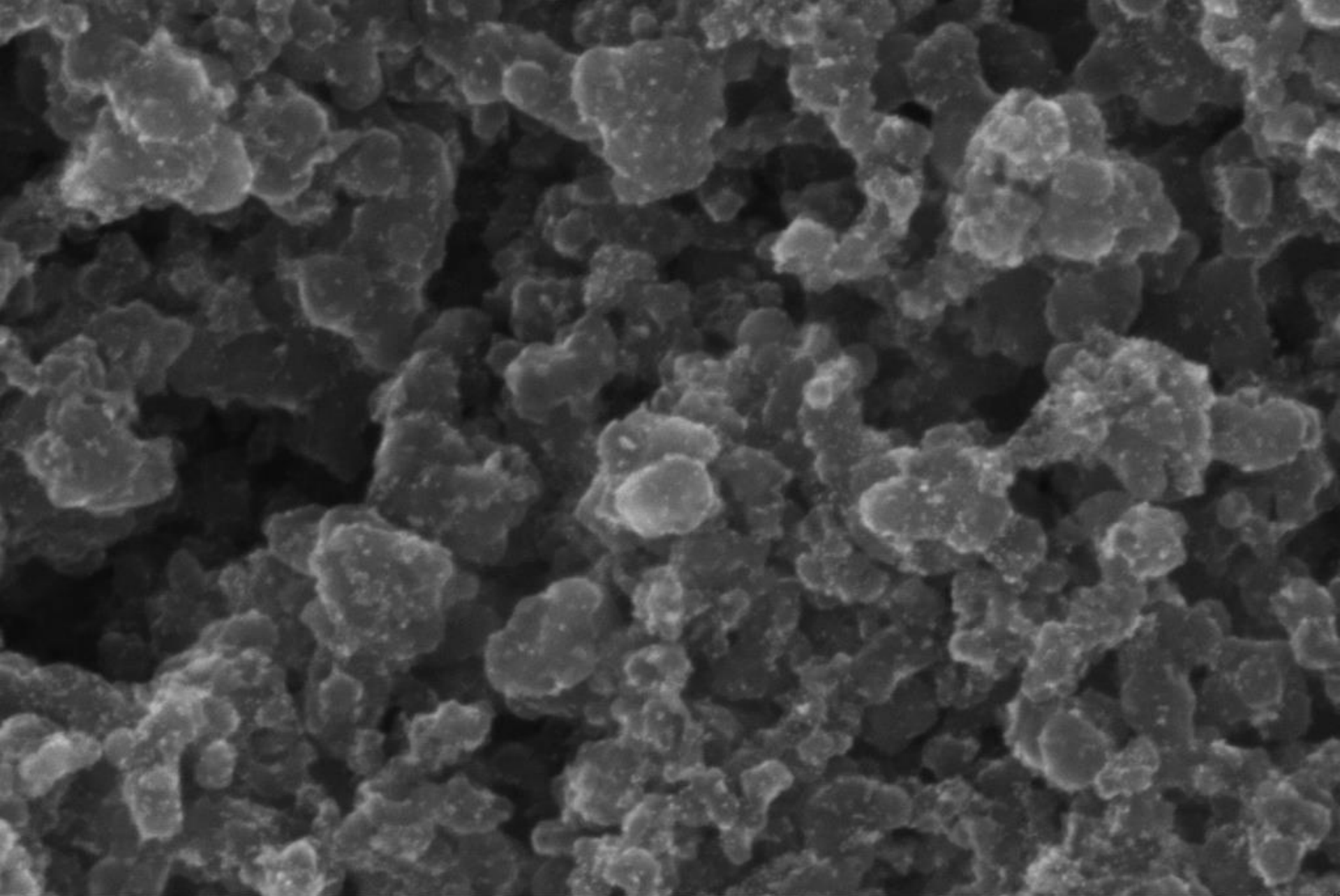
100 nm



EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.85e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μ m File Name = Denora_Pt_27.tif

Date : 23 Oct 2012





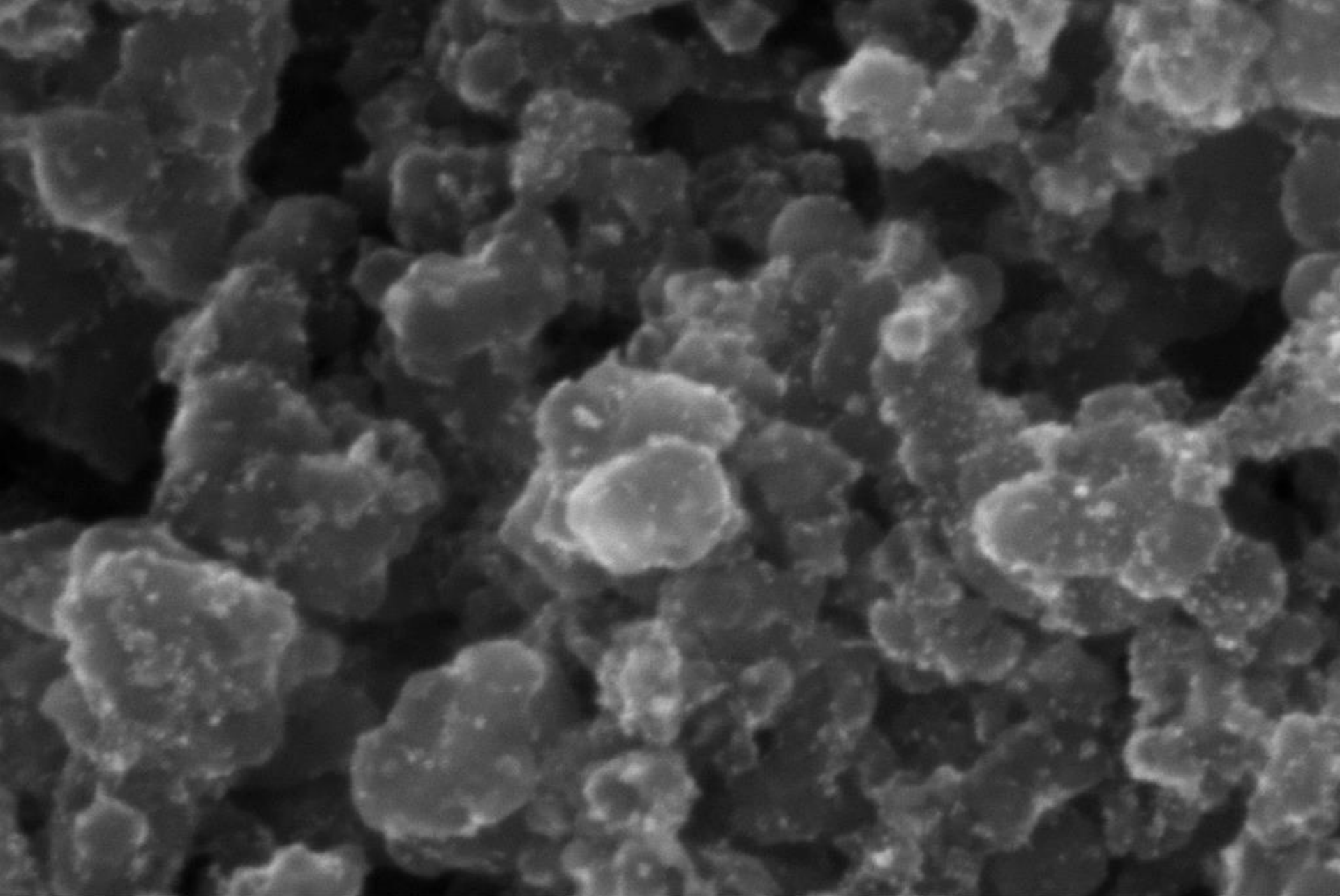
100 nm



EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.87e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μ m File Name = Denora_Pt_26.tif

Date : 23 Oct 2012





100 nm

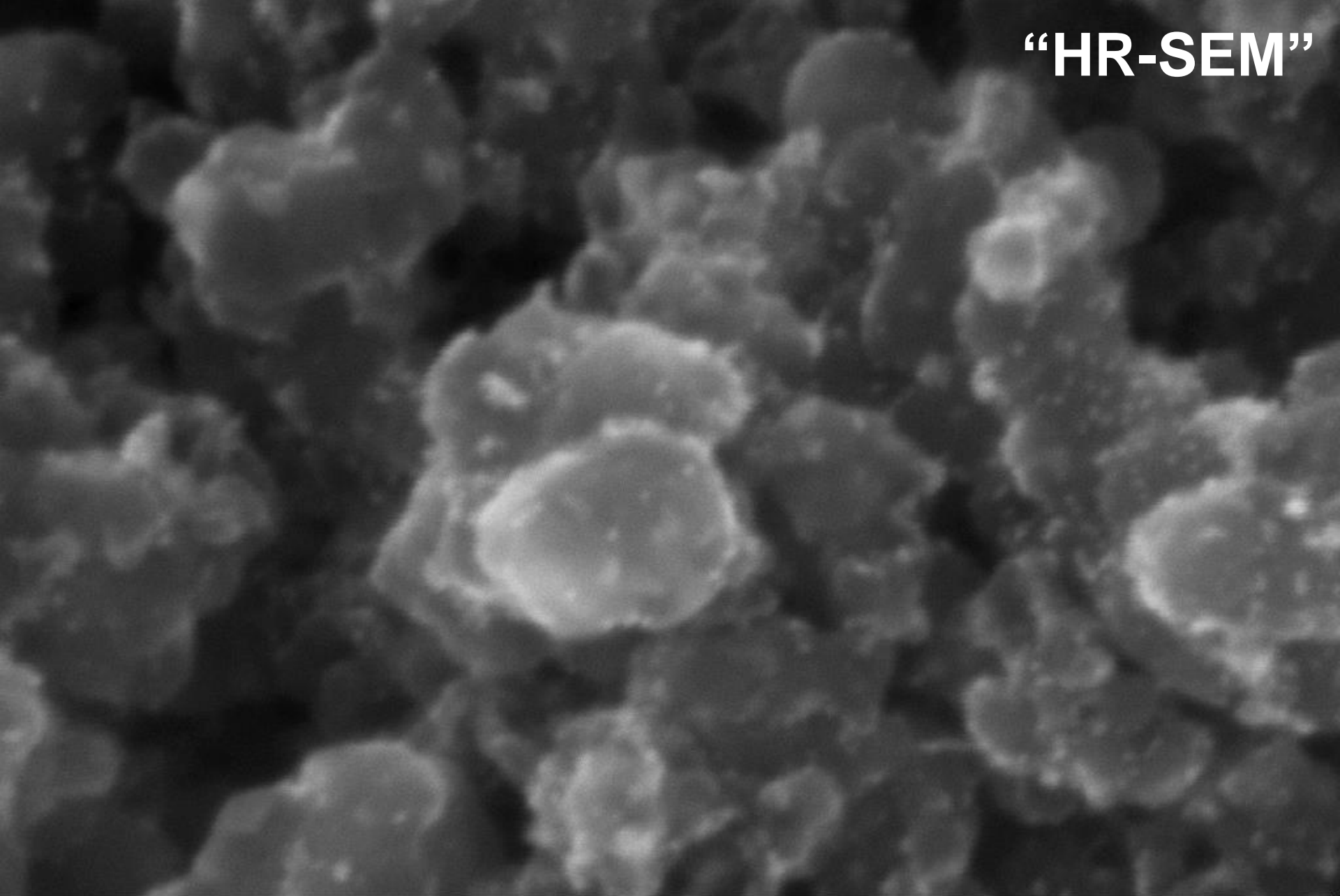


EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.87e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μ m File Name = Denora_Pt_25.tif

Date : 23 Oct 2012



“HR-SEM”



20 nm

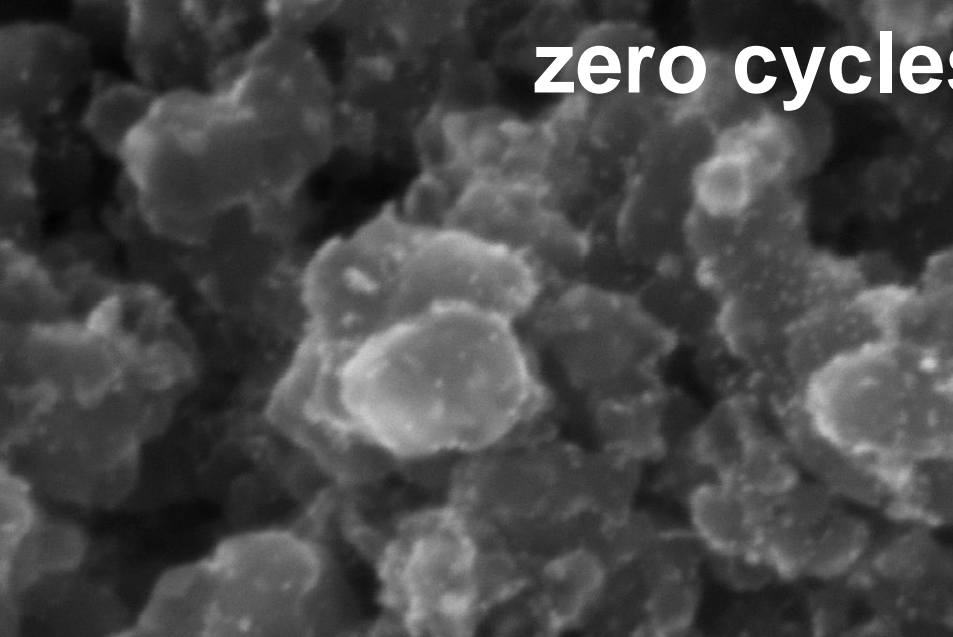


EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.85e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μ m File Name = Denora_Pt_24.tif

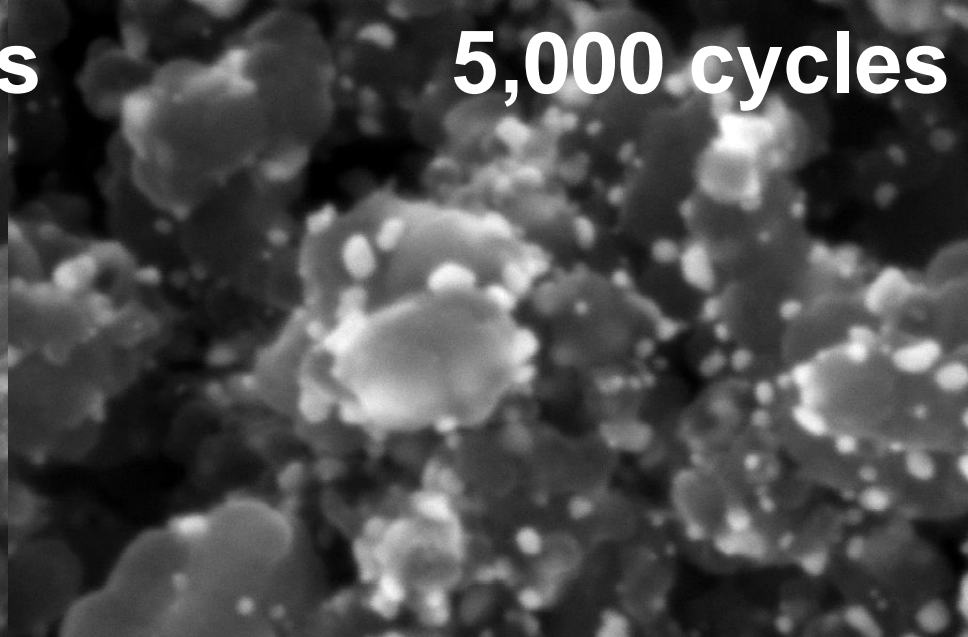
Date : 23 Oct 2012



zero cycles



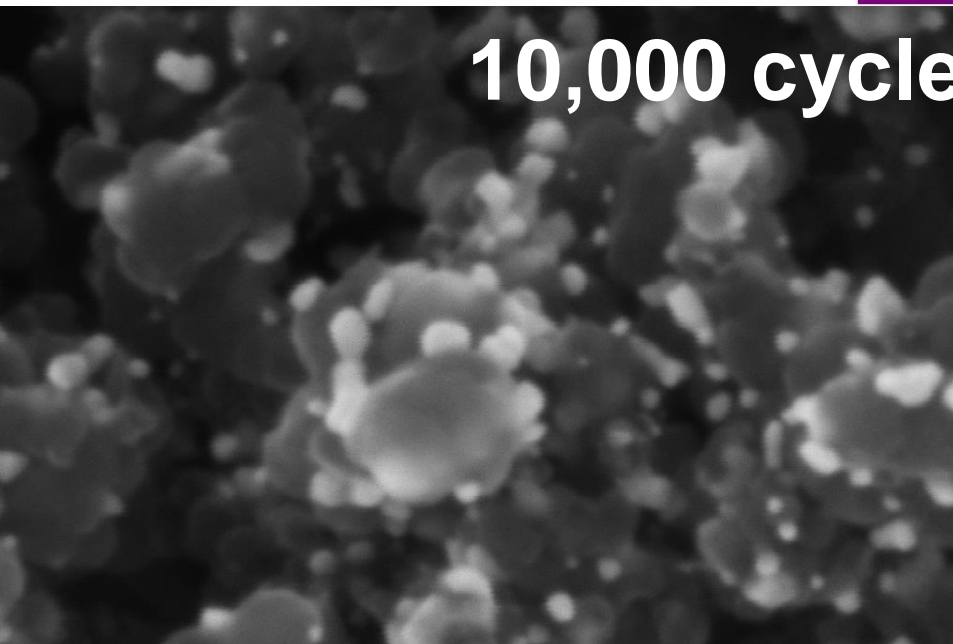
5,000 cycles



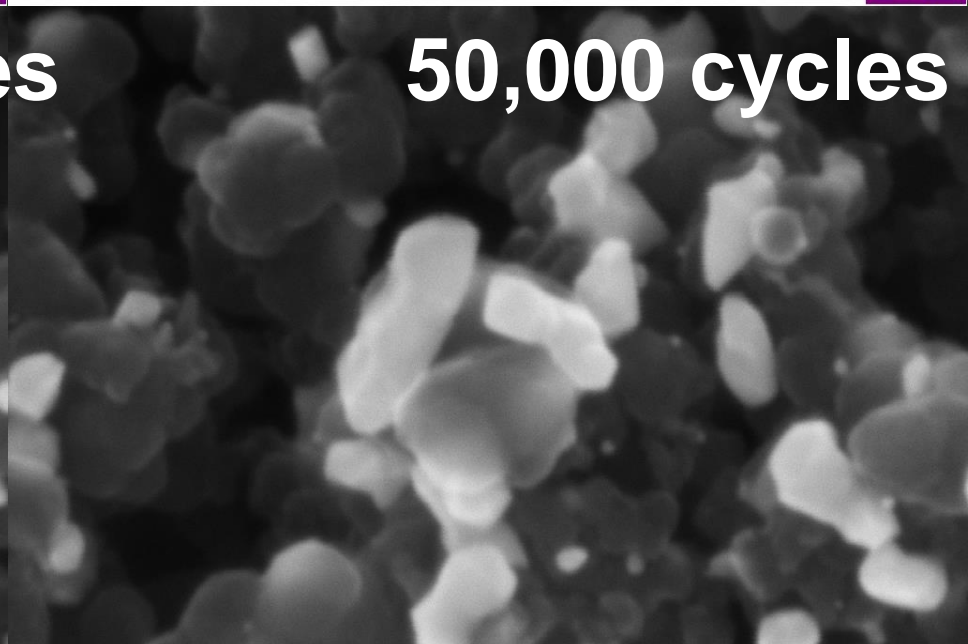
20 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.85e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μ m File Name = Denora_Pt_24.tif
Date :23 Oct 2012

20 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0300 Chamber = 6.60e-004 Pa
WD = 4.7 mm Aperture Size = 10.00 μ m File Name = Denora-Pt_5000_13.tif
Date :24 Oct 2012

10,000 cycles



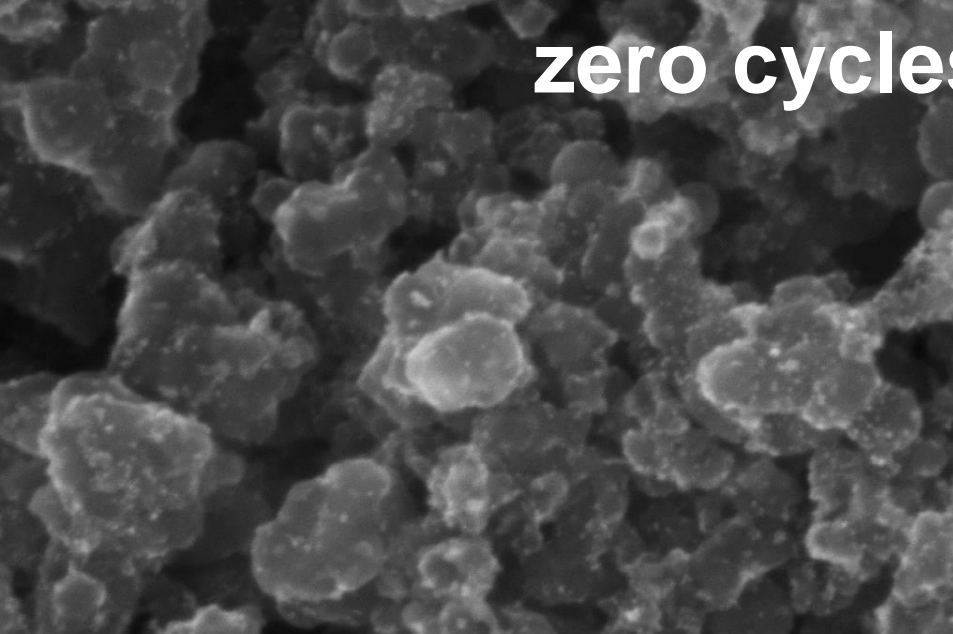
50,000 cycles



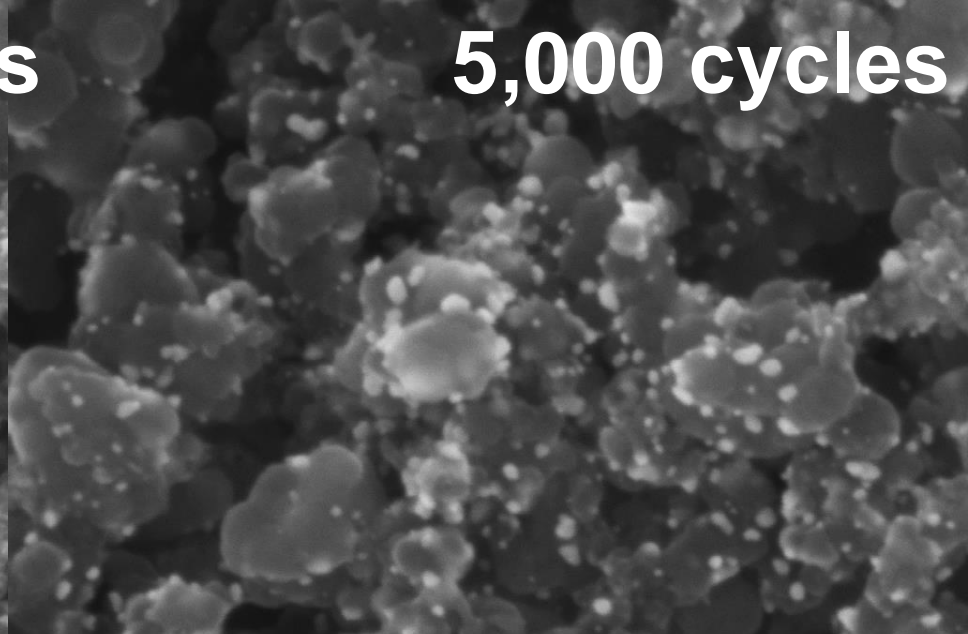
20 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0300 Chamber = 7.27e-004 Pa
WD = 4.7 mm Aperture Size = 10.00 μ m File Name = denora_Pt_10000_11.tif
Date :25 Oct 2012

20 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0200 Chamber = 6.68e-004 Pa
WD = 4.7 mm Aperture Size = 10.00 μ m File Name = denora_Pt_50000_12.tif
Date :29 Oct 2012

zero cycles



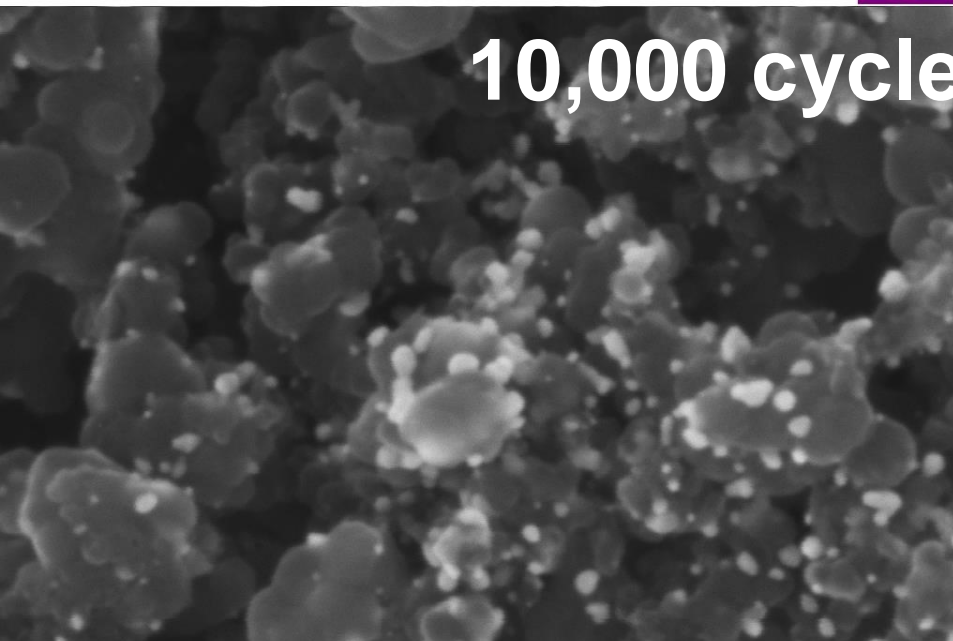
5,000 cycles



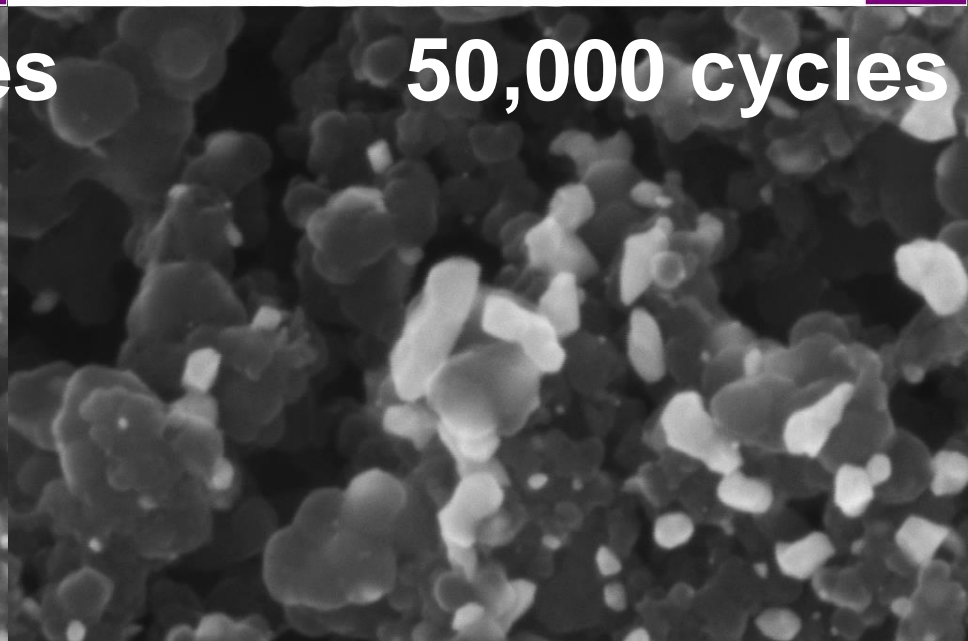
100 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.87e-004 Pa Date :23 Oct 2012
WD = 4.8 mm Aperture Size = 10.00 µm File Name = Denora_Pt_25.tif

100 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0300 Chamber = 6.49e-004 Pa Date :24 Oct 2012
WD = 4.7 mm Aperture Size = 10.00 µm File Name = Denora-Pt_5000_14.tif

10,000 cycles



50,000 cycles



100 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0300 Chamber = 7.15e-004 Pa Date :25 Oct 2012
WD = 4.7 mm Aperture Size = 10.00 µm File Name = denora_Pt_10000_12.tif

100 nm
EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0200 Chamber = 6.49e-004 Pa Date :29 Oct 2012
WD = 4.7 mm Aperture Size = 10.00 µm File Name = denora_Pt_50000_13.tif

zero cycles

5,000 cycles

100 nm

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0400 Chamber = 3.87e-004 Pa
WD = 4.8 mm Aperture Size = 10.00 μ m File Name = Denora_Pt_26.tif

Date :23 Oct 2012



100 nm

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0300 Chamber = 6.35e-004 Pa
WD = 4.7 mm Aperture Size = 10.00 μ m File Name = Denora-Pt_5000_15.tif

Date :24 Oct 2012



10,000 cycles

50,000 cycles

100 nm

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0300 Chamber = 6.99e-004 Pa
WD = 4.7 mm Aperture Size = 10.00 μ m File Name = denora_Pt_10000_13.tif

Date :25 Oct 2012



100 nm

EHT = 7.00 kV Signal A = InLens Mix Signal = 0.0200 Chamber = 6.49e-004 Pa
WD = 4.7 mm Aperture Size = 10.00 μ m File Name = denora_Pt_50000_14.tif

Date :29 Oct 2012



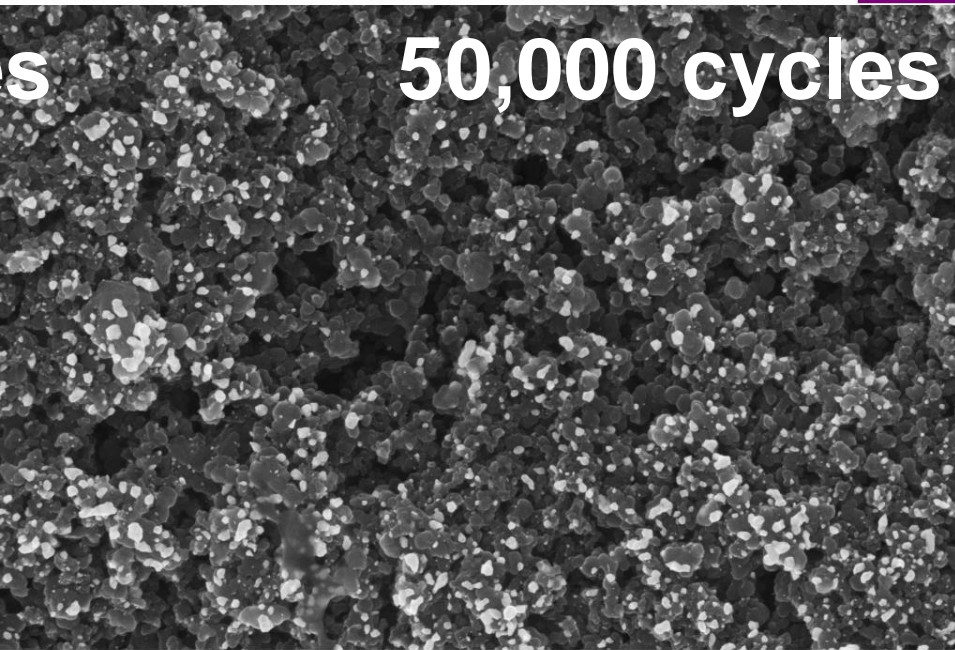
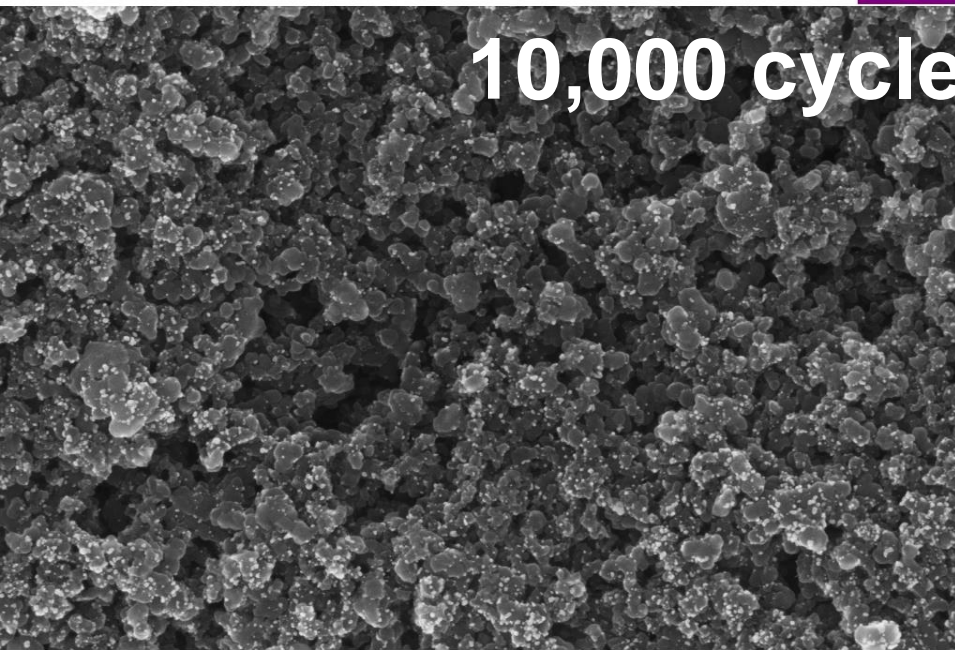
zero cycles

5,000 cycles



10,000 cycles

50,000 cycles



zero cycles

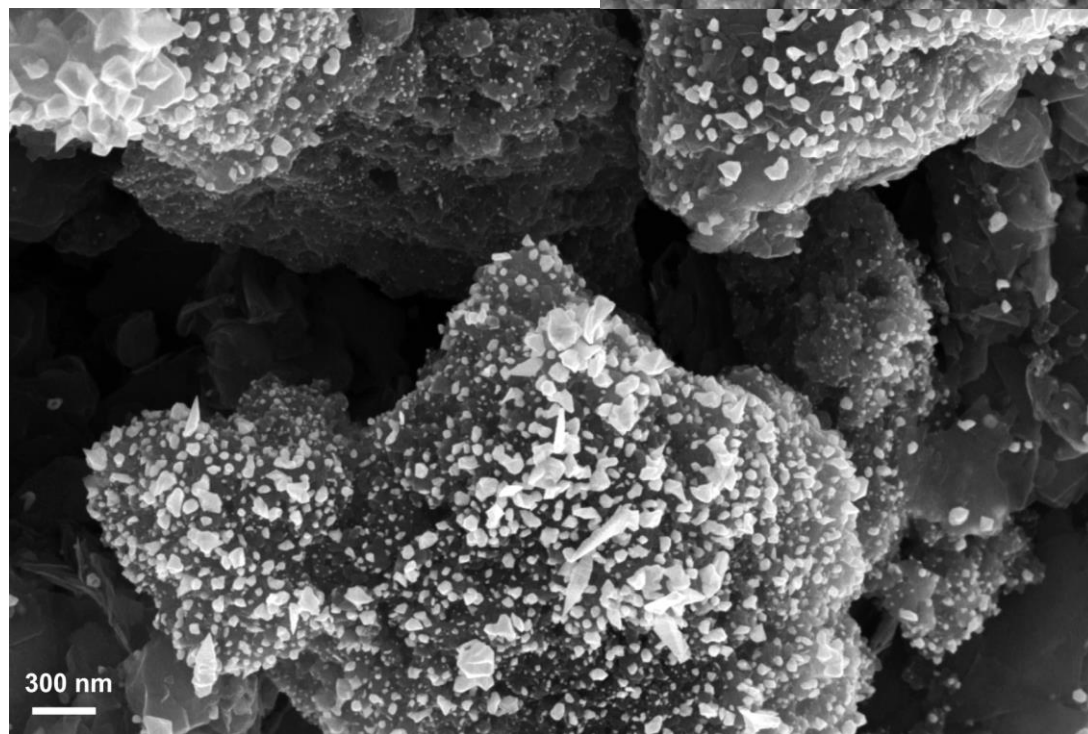
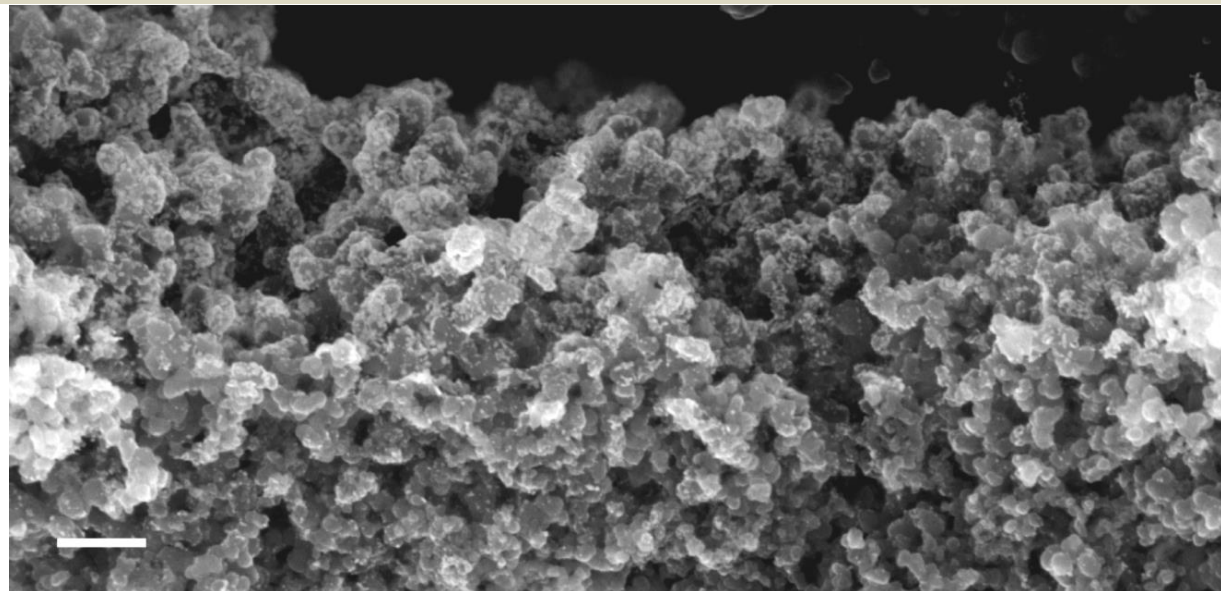
5,000 cycles



10,000 cycles

50,000 cycles





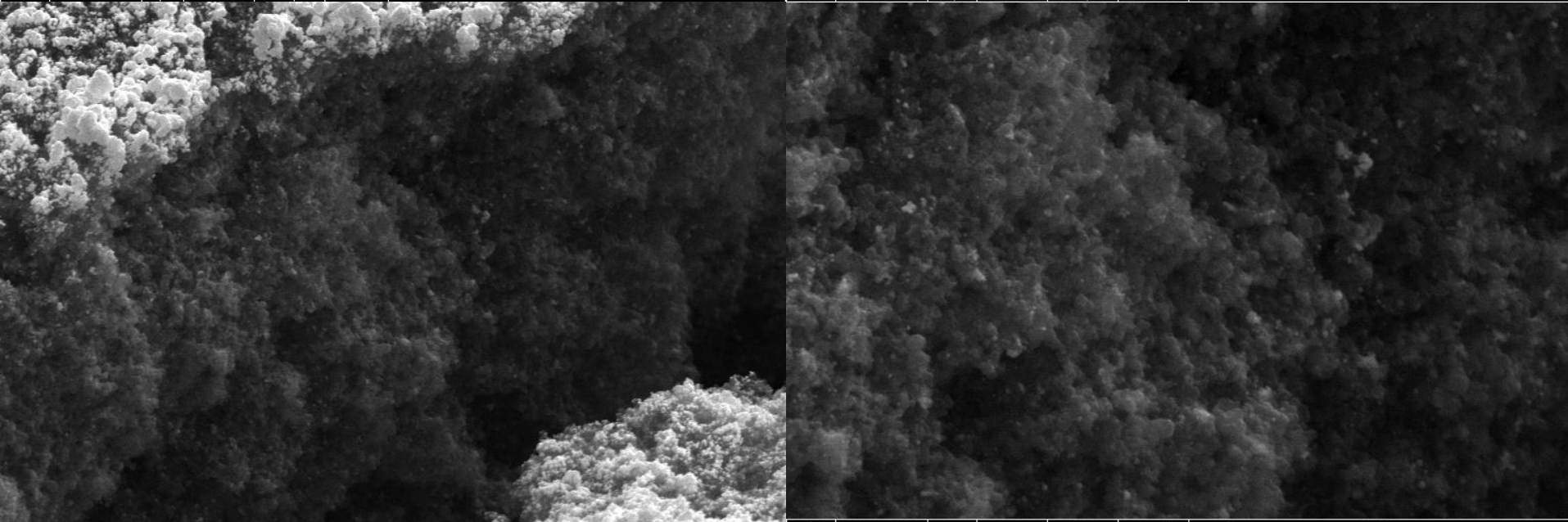


	11/12/2012 3:27:00 PM	dwell 3 μs	HV 10.00 kV	HFWD 41.4 μm	mag 5 000 ×
--	--------------------------	---------------	----------------	-----------------	----------------

10 μm	
Helios	

	11/12/2012 3:15:54 PM	dwell 12 μs	HV 10.00 kV	HFWD 20.7 μm	mag 10 000 ×
--	--------------------------	----------------	----------------	-----------------	-----------------

5 μm	
Helios	

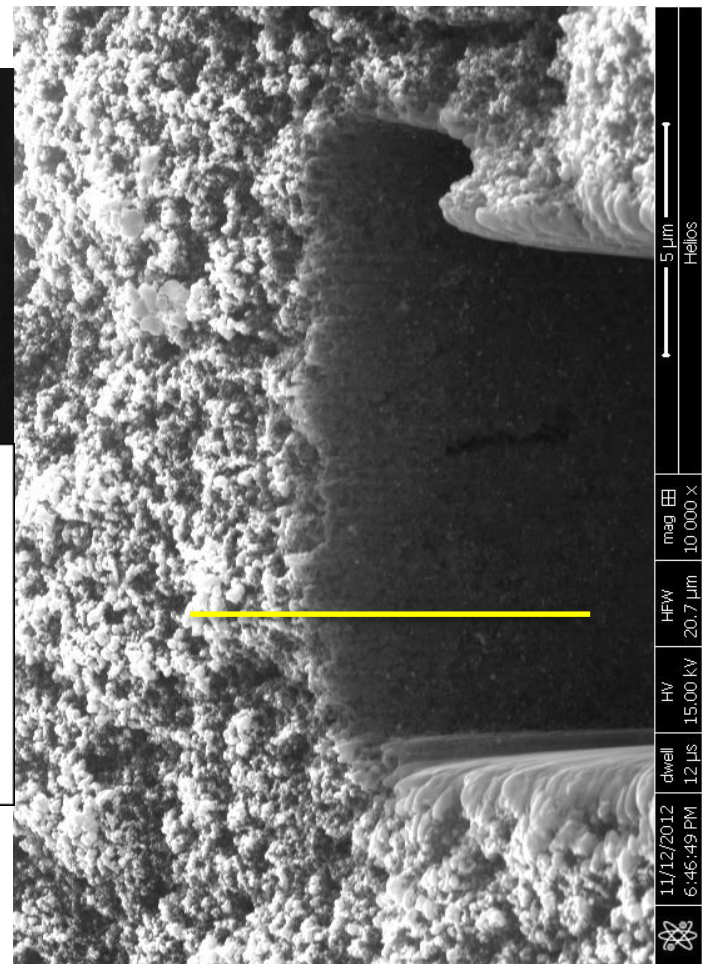
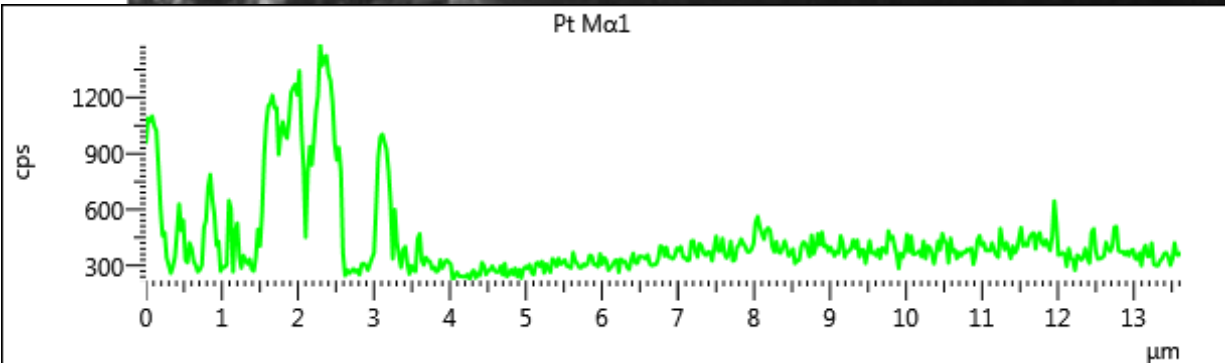
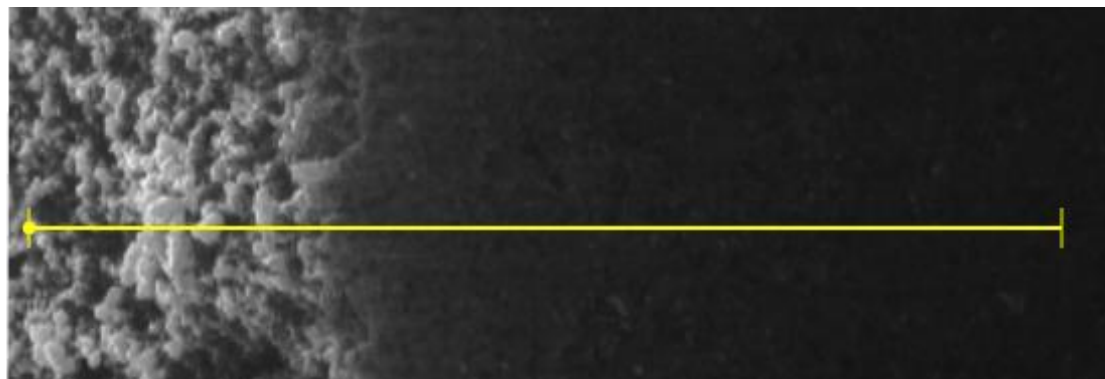


	11/12/2012 3:14:34 PM	dwell 12 μs	HV 10.00 kV	HFWD 10.4 μm	mag 20 000 ×
--	--------------------------	----------------	----------------	-----------------	-----------------

4 μm	
Helios	

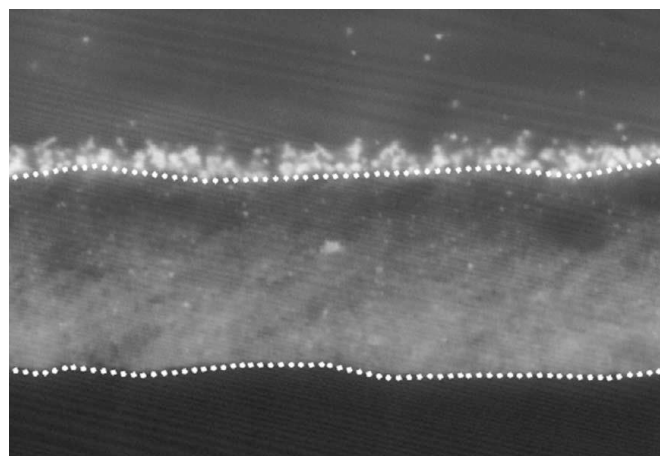
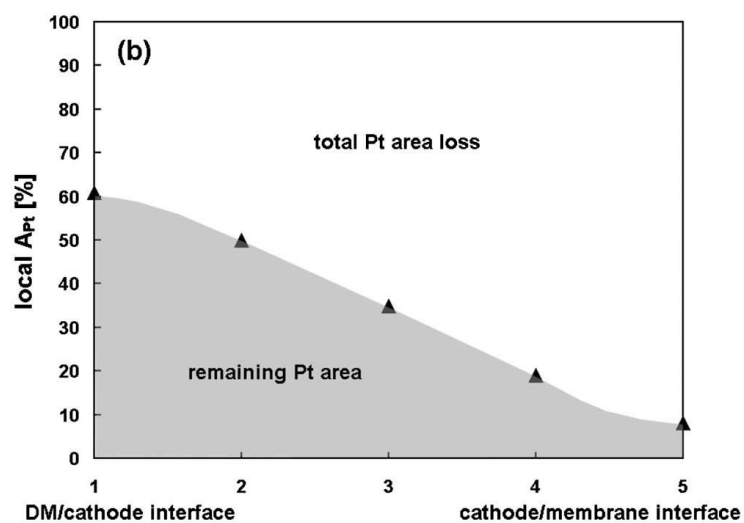
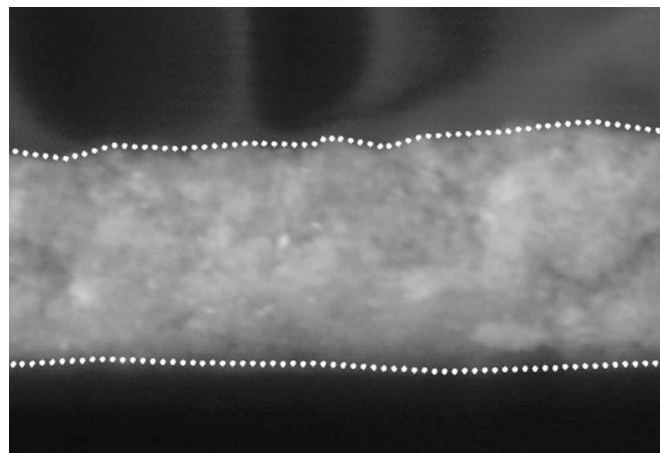
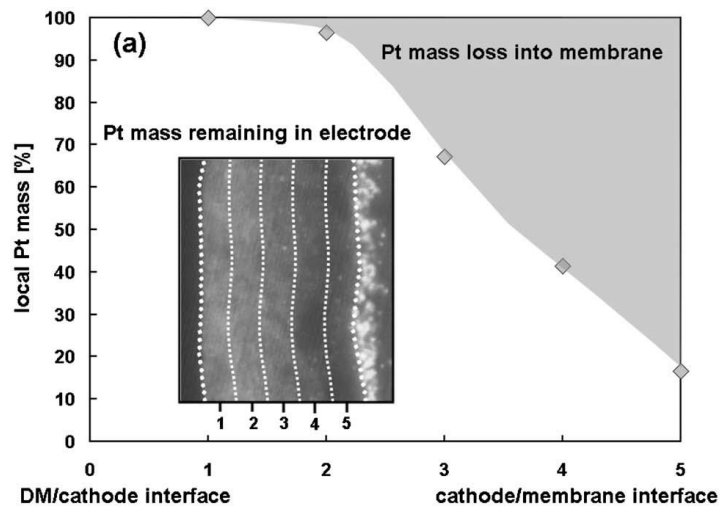
	11/12/2012 3:14:05 PM	dwell 12 μs	HV 10.00 kV	HFWD 5.18 μm	mag 40 000 ×
--	--------------------------	----------------	----------------	-----------------	-----------------

2 μm	
Helios	



11/12/2012	dwell	12 μ s	HV	15.00 kV	FW	20.7 μ m	mag	10 000 X
6:46:49 PM								

2010_Gasteiger_Shao_Platinum-Alloy Cathode Catalyst Degradation in Proton Exchange Membrane Fuel Cells: Nanometer-Scale Compositional and Morphological Changes



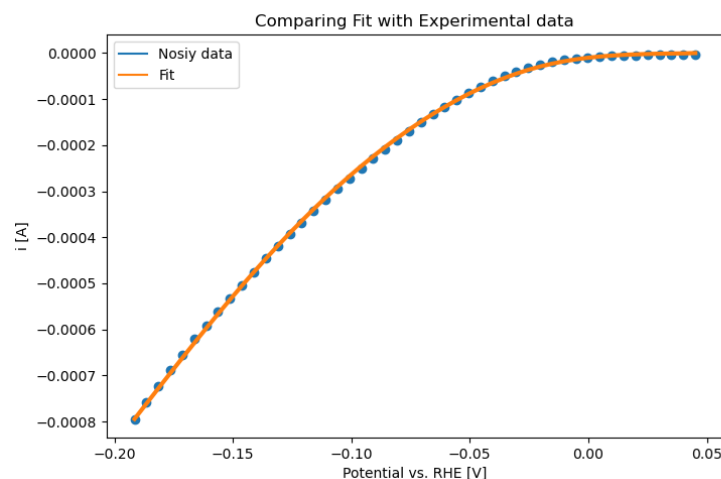
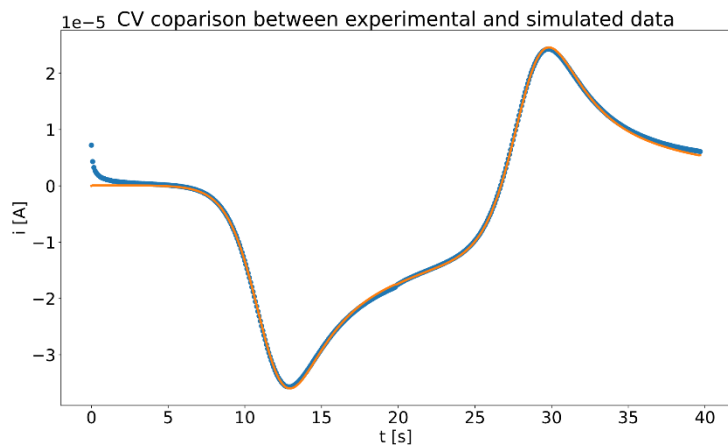
What do we do in the lab on these topics?

- Computer vision for TEM images
- Pattern recognition for electron diffractions
- Electrochemistry data analysis/simulation
- Automatization of gas diffusion electrode measurement
- Compositionally complex alloys - screening
 - data collection in a readable database; “electronic lab notebook” – Quipnex (our own FAIRmat)

Inferring kinetic parameters from experiments

- Finding optimal parameters:
 - An objective approach to comparing materials
 - Finding and fitting a mechanism onto experimental data using ML approaches

Comparing HER data with simulation

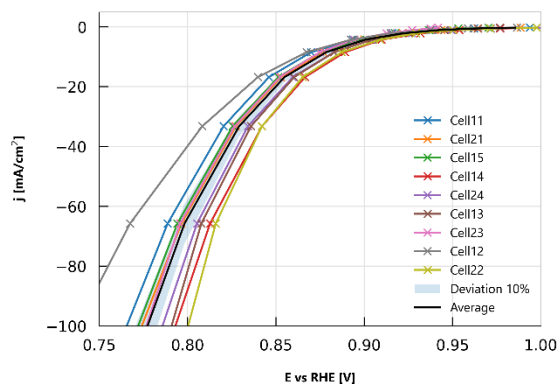
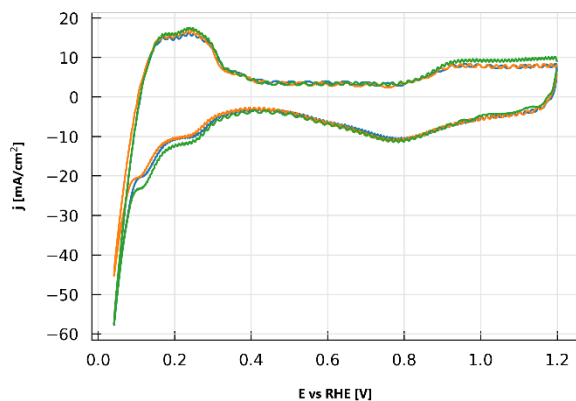
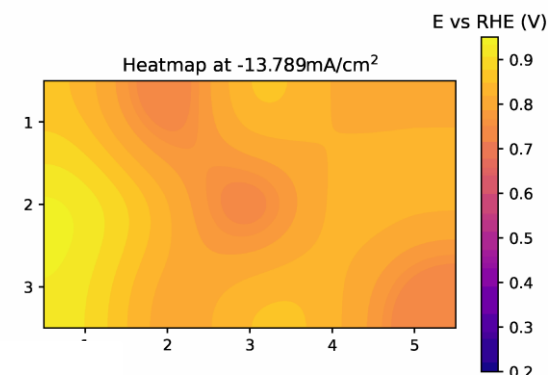


Thanks Ožbej Vodeb!

From Data to Discovery: Designing a Relational Electrocatalyst Database for Machine Learning-Driven Pattern Recognition



- We have begun experimenting with high-throughput measurements – automatization
- To test reproducibility – error bars
- There is a lot of data being generated with the measurements
- It needs to be stored somewhere
- A unified place where data is being gathered (and processed)



Thanks Aleš Arsel and Miha Hotko!

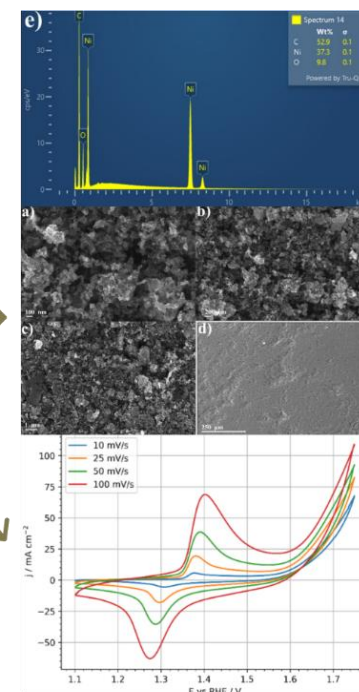
From Data to Discovery: Designing a Relational Electrocatalyst Database for Machine Learning-Driven Pattern Recognition



- Database tables have key descriptors to store and organize experimental data effectively
There has to exist a sample table which every other table references
- The interface needs to have an API for the programmatic uploading of data
- Measurements still require a human to determine the reliability of the generated result - **a Trust Factor**
- There needs to be a way to determine why the results aren't trusted
- This data could be used for machine learning once the database has enough information
- Good care has to be taken in designing the database schema.

Catalyst synthesis

Catalyst characterization



Thanks Aleš Arsel and Miha Hotko!

Modifications of compositionally complex materials

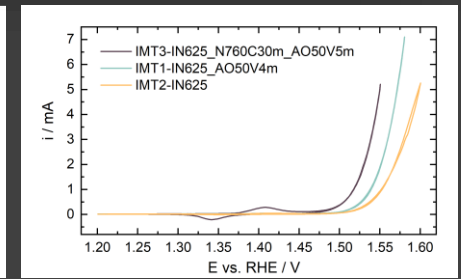
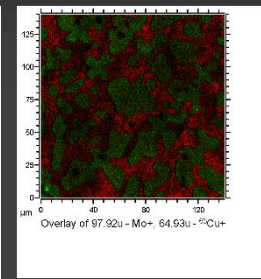
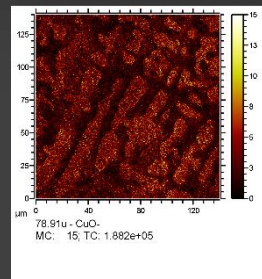
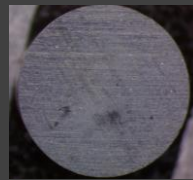
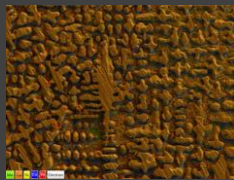
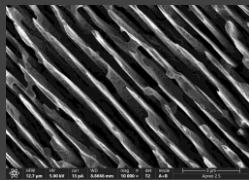
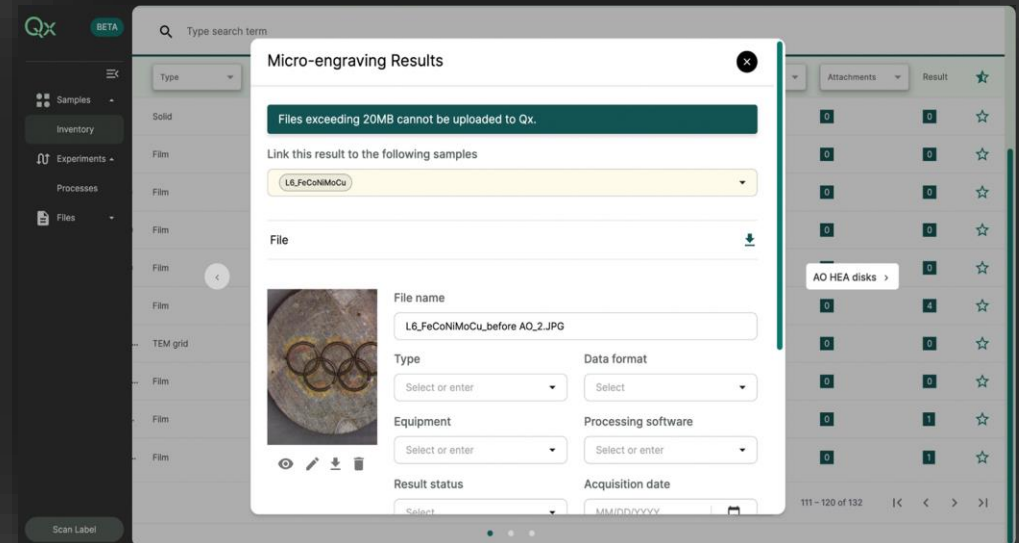
Various alloys

▼
Electrochemical treatments

Heat treatments

▼
Analyses

▼
Scale up



Thanks Luka Suhadolnik!

Data organization – the Qx app

OER activity
Last edit: 14-05-2024, 16:48

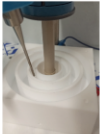
Samples

Apply process to: 1
Measurement date: 13-05-2024

Electrochemical cell

Cleaning procedure: Boiling for 2 h
Material: PTFE

Photo of the electrochemical cell



Electrolyte

Volume: 25 mL
Type: 1 M KOH Titripur
Preparation date / R...: 18-04-2024

Electrolyte usage count / No. of ...: 1
Electrolyte status after the mea...: Wasted

CV

Potentiostat: BioLogic

Potential range: 1.2 - 1.55 mV

Scan rate: _____
Number of cycles: _____



- Hierarchical sample creation and linkage
- Adding of processes, comments, and results
- Creation of process templates with all the important parameters
- Exports for advanced analysis

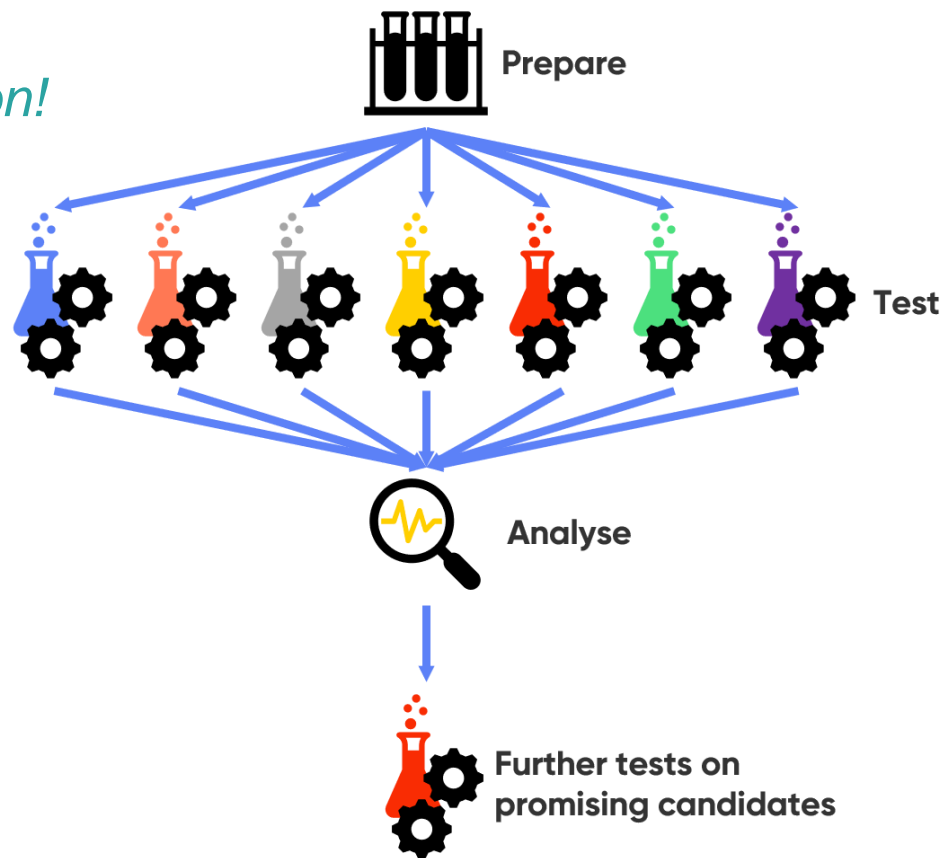
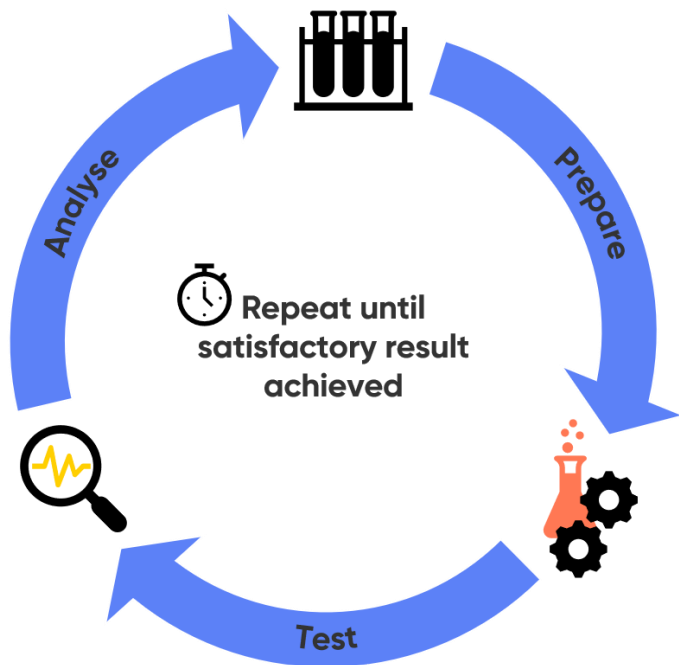
<https://www.quipnex.com>

Thanks Luka Suhadolnik!

Our wish! Jubilee for electrocatalysis!

Or, more realistically, just automatization of some of our experiments...

High throughput screening → solution!



Problems?
 Reliable measurements
 Reliable data collection
 Data processing
 Complexity?
 Materials discovery?
 Black box?
 We still try!

Thank you for your attention!

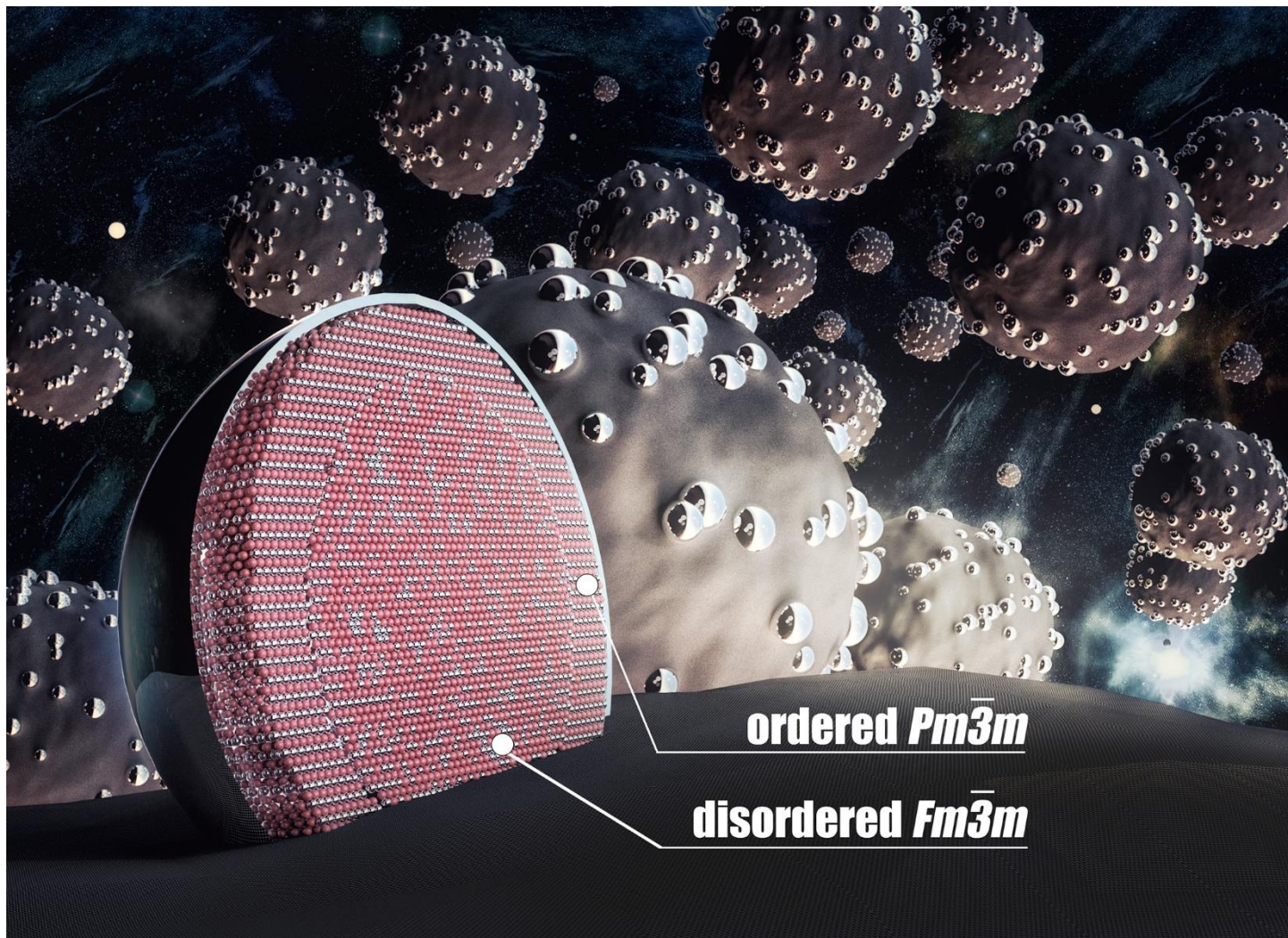


RESEARCH & INNOVATION PROGRAMME
ON RAW MATERIALS
TO FOSTER CIRCULAR ECONOMY

“Besides improving the technology we should also try to adapt human habits.”

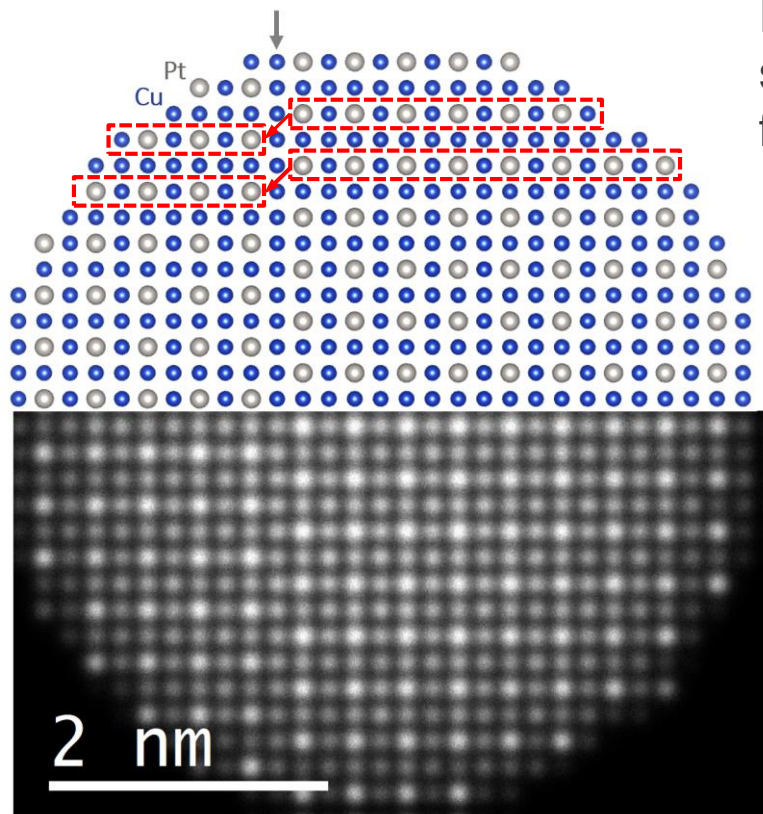
The devil is in the details.

Structurally ordered Pt-alloy nanoparticles (~2010)

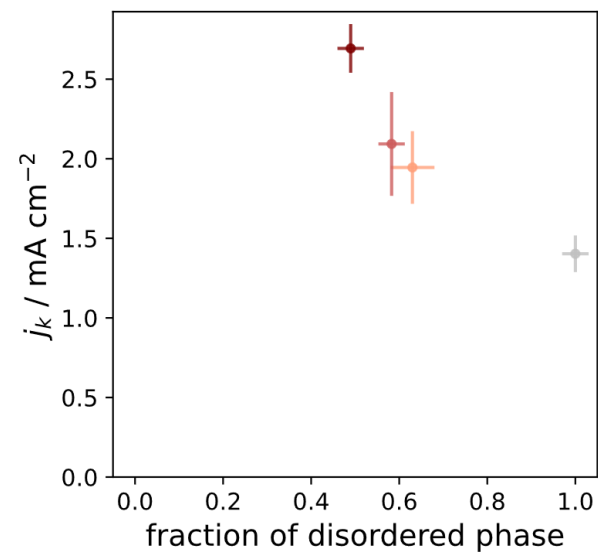


The devil is in the details.

Pt-alloy structure

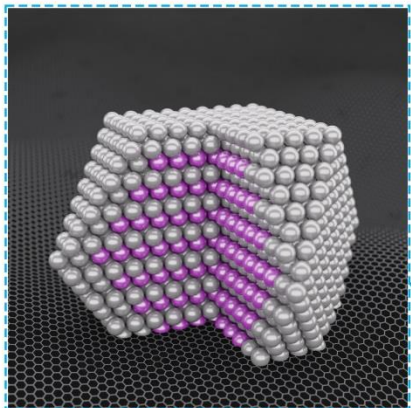


Periodic anti-phase boundaries and crystal superstructures in PtCu₃ nanoparticles as fuel cell catalysts

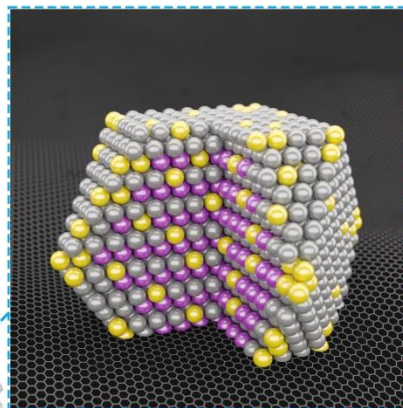


Short facts of innovative technology

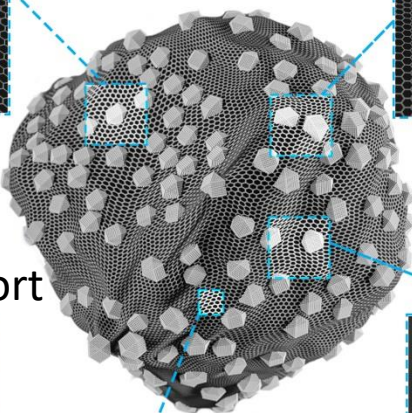
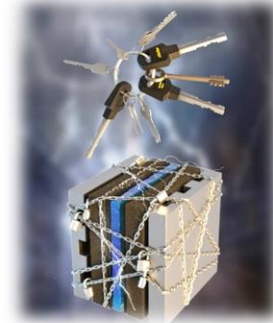
Crystal ordering



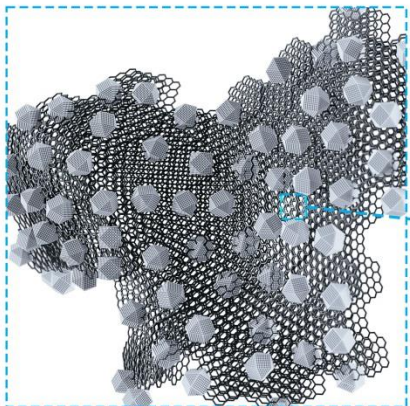
Crystal doping



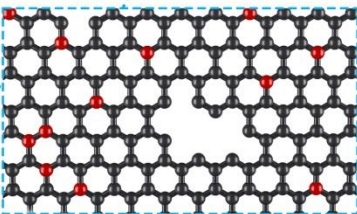
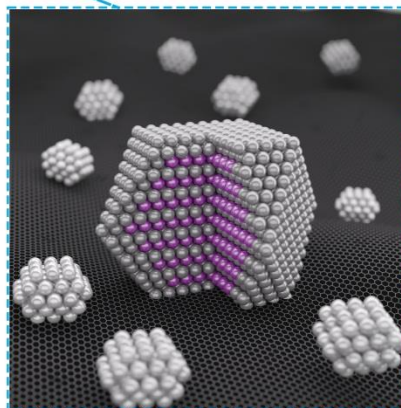
“Unlocking potentials”



Graphene-based support



Decoration



Support doping

KI startups:

MEAs:



mebius

Catalysts:



Contact: nejc.hodnik@ki.si