



Machine Learning for Investigation of Nickel Surface Chemistry in Electrocatalytic Production of Hydrogen

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Ljubljana, May 2024

ARIS Funded Fundamental project

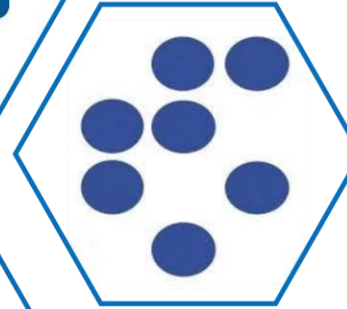
Fundamental understanding of Hydrogen Evolution Reaction for a new generation of nickel-based electrocatalysts in alkaline water and chlor-alkali electrolysis

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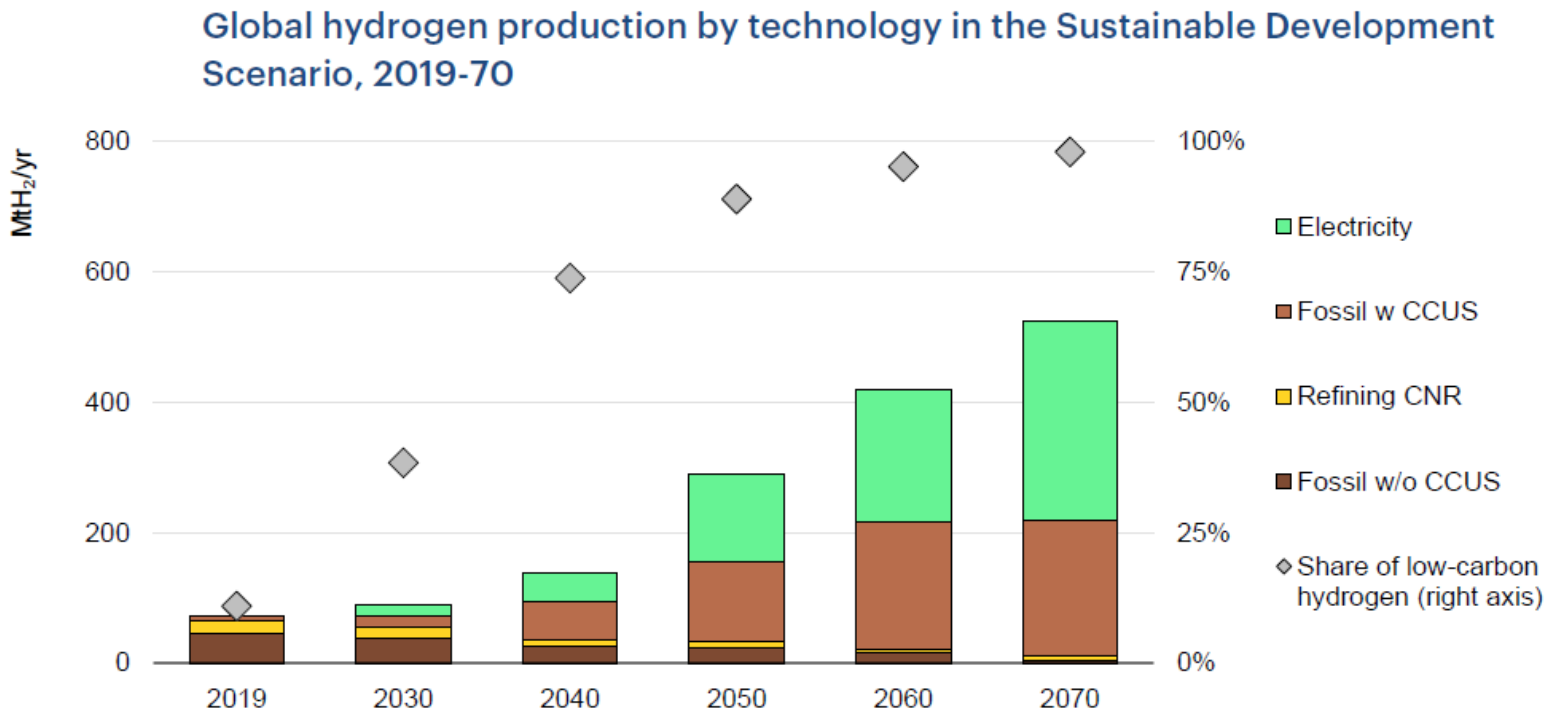
aris



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Motivation - IEA – Energy technology perspectives



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Note: CNR = hydrogen as by-product from catalytic naphtha reforming in refineries.

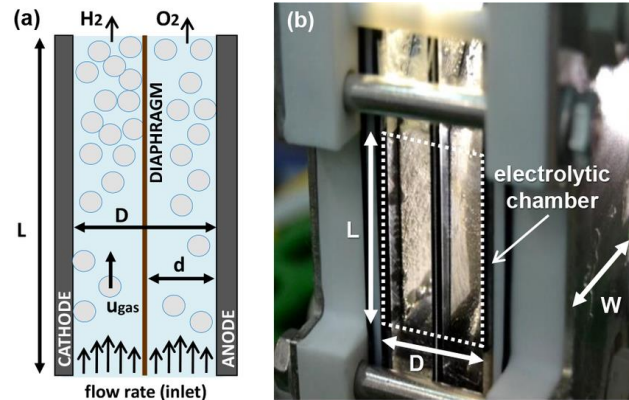
Today, hydrogen is produced almost entirely from fossil fuels without CCUS. Low-carbon hydrogen production dominates in the Sustainable Development Scenario, with almost all hydrogen production either from low-carbon electricity or fossil fuels with CCUS.

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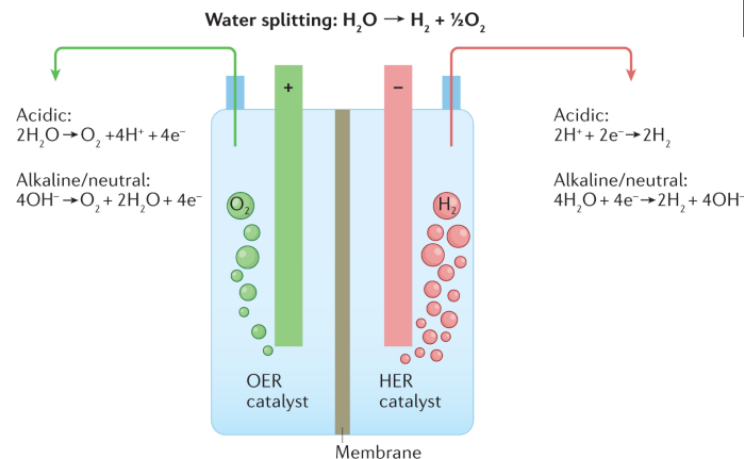
Water electrolysis



Small alkaline water electrolyzer stack



Alkaline water electrolyzer single cell



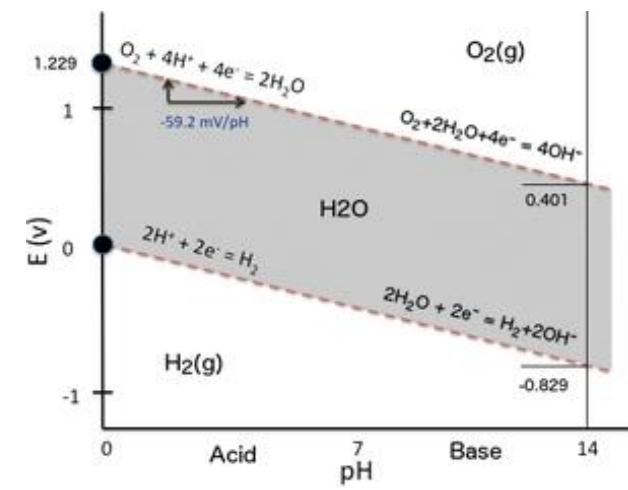
Shih, A.J., Monteiro, M.C.O., Dattila, F. *et al.* Water electrolysis. *Nat Rev Methods Primers* **2**, 84 (2022).

$2\text{H}_2\text{O} \rightarrow 2\text{H}_2 + \text{O}_2 \quad \Delta G = -nFE_{\text{cell}}$

$\Delta G_{\text{H}_2\text{O}} = -237 \text{ kJ/mol} \rightarrow E_{\text{cell}} = 1.23 \text{ V}$

Cathodic branch:
 $4\text{H}_2\text{O} + 4\text{e}^- \rightarrow 2\text{H}_2 + 4\text{OH}^-$

Anodic branch:
 $2\text{H}_2\text{O} \rightarrow \text{O}_2 + 4\text{H}^+ + 4\text{e}^-$



Objective(s)

This proposal aims to advance the fundamental understanding of HER on nickel-based cathode catalysts and to utilize this knowledge to develop a new generation of nickel catalysts with 10-times higher activity than state of the art.

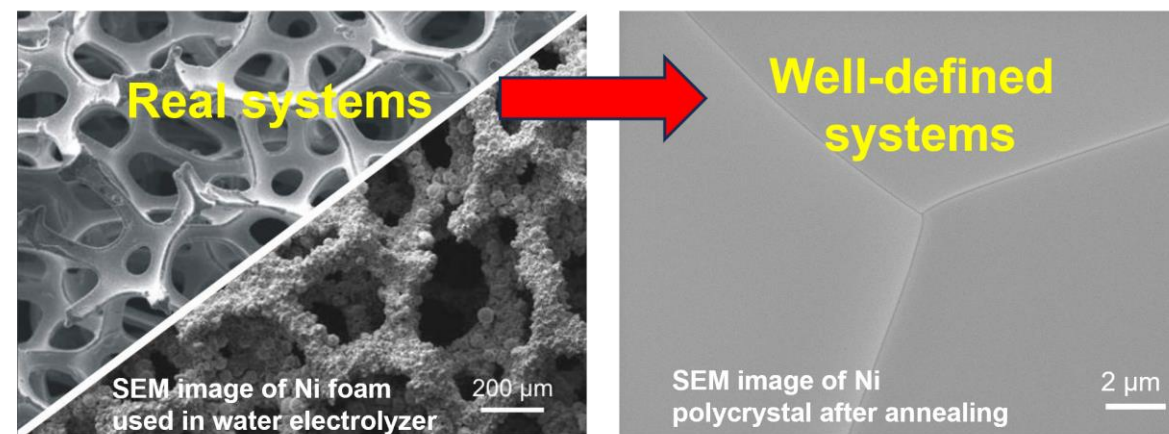
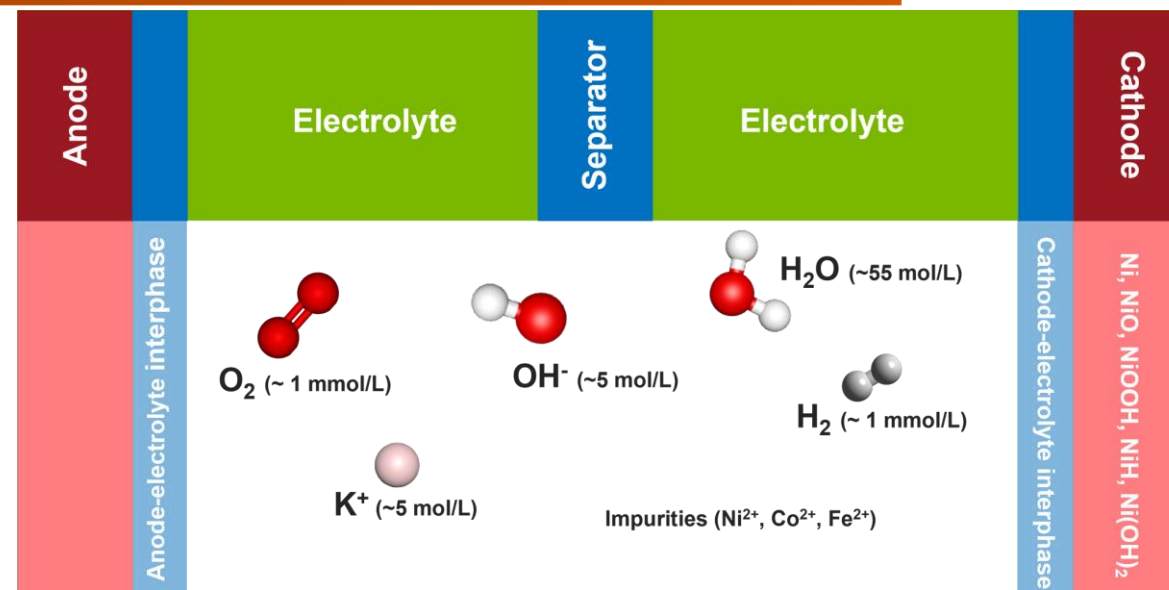
- 1) What are the active sites for HER on Ni catalysts?
- 2) What is the role of individual Ni surface species (Ni^0 , NiO , Ni(OH)_2 , Ni(OOH) , etc.) in HER electrocatalysis?
- 3) What is the role of electrolyte pH, cations and anions in HER electrocatalysis on Ni?

Can we find a way to control, influence and alter the electrochemical interface and its properties, instead of just observing and evaluating different combinations of the electrode materials and electrolytes?



Approach

- The spread of literature-reported catalytic activities for HER on Ni is quite large – a sign of complexity of the system stemming from rich Ni surface chemistry
- Simplify complex Ni-based systems to well-defined systems
- Develop/utilize several Ni platforms, that connect model to real systems
- Develop new tools and methods for the investigation of electrode-electrolyte interphase
- Execute detailed analysis of Ni-based materials and electrolyte using established analytical tools and methods (EIS, ToF-SIMS, FTIR, Raman, XPS, RRDE, AFM)
- Develop understanding how individual components of the cathode-electrolyte interphase behave at potentials relevant for water electrolysis
- Determine intrinsic nickel electrochemical properties over many different nickel platforms, from single atom, over nanoparticles to extended, micro and millimeter size samples.
- Develop next generation of materials for HER in alkaline water and chlor-alkali electrolysis

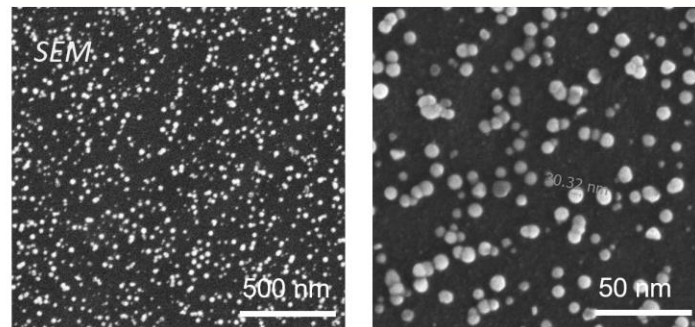


Approach

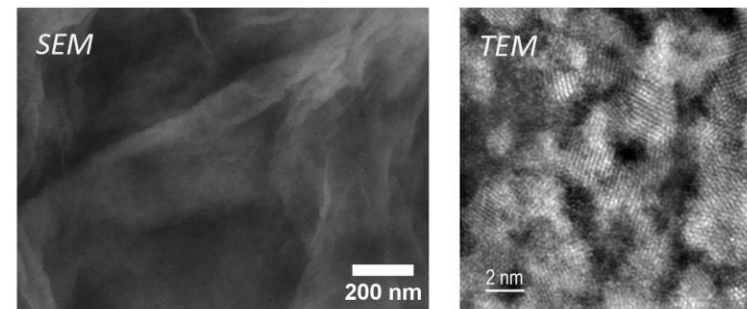
Synthesis of Ni platforms

1. Nickel polycrystal
2. Nickel based nanoparticles (chemically synthesized)
3. Nickel nanoparticles (electrodeposited on GC)
4. Ni TEM grid
5. Ni nano-wires
6. Ni single crystals

Nickel nanoparticles (electrodeposited)

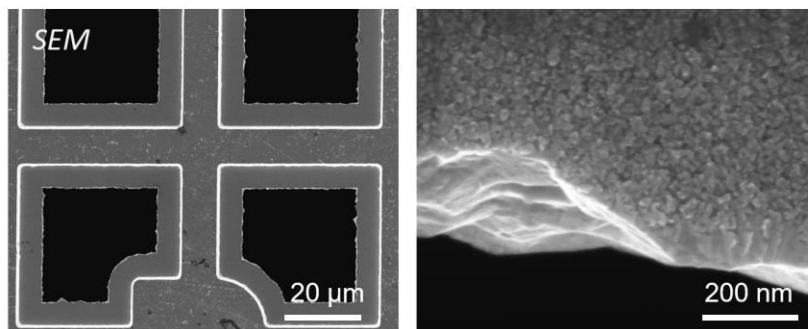


Nickel-based nanoparticles (chemically synthesized)

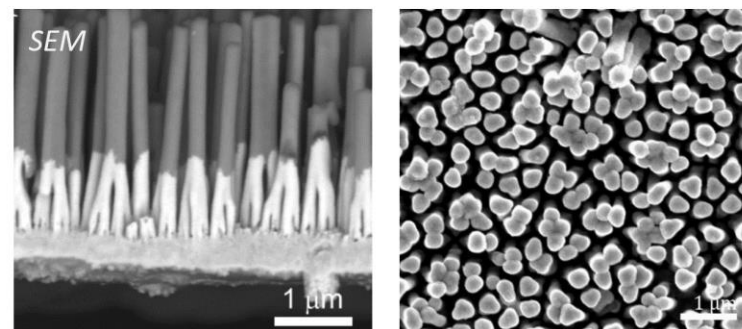


2D Nickel hydroxide on rGO

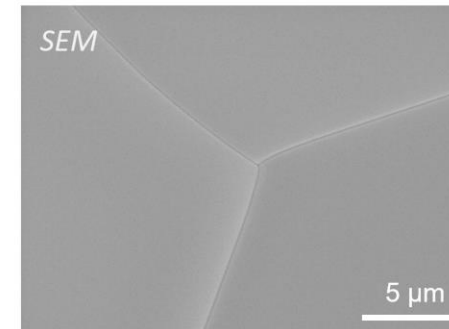
Ni TEM grid (unmodified)



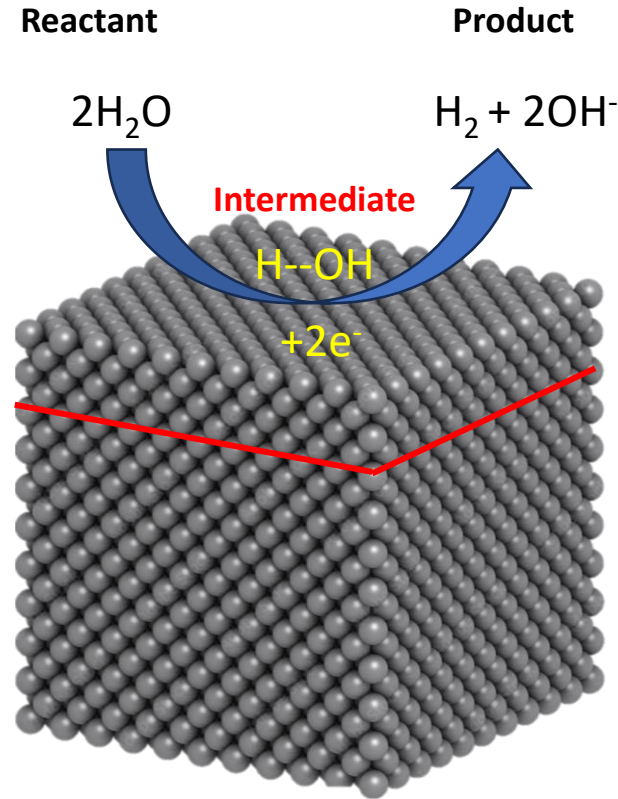
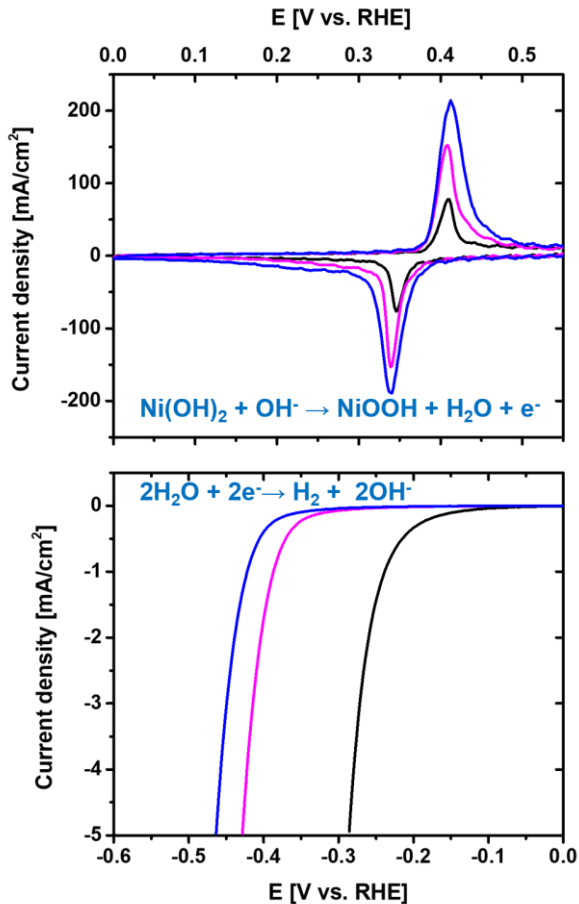
Ni nano-wires (grown using AAO template)



Polycrystalline Ni disk (annealed at 1000°C)

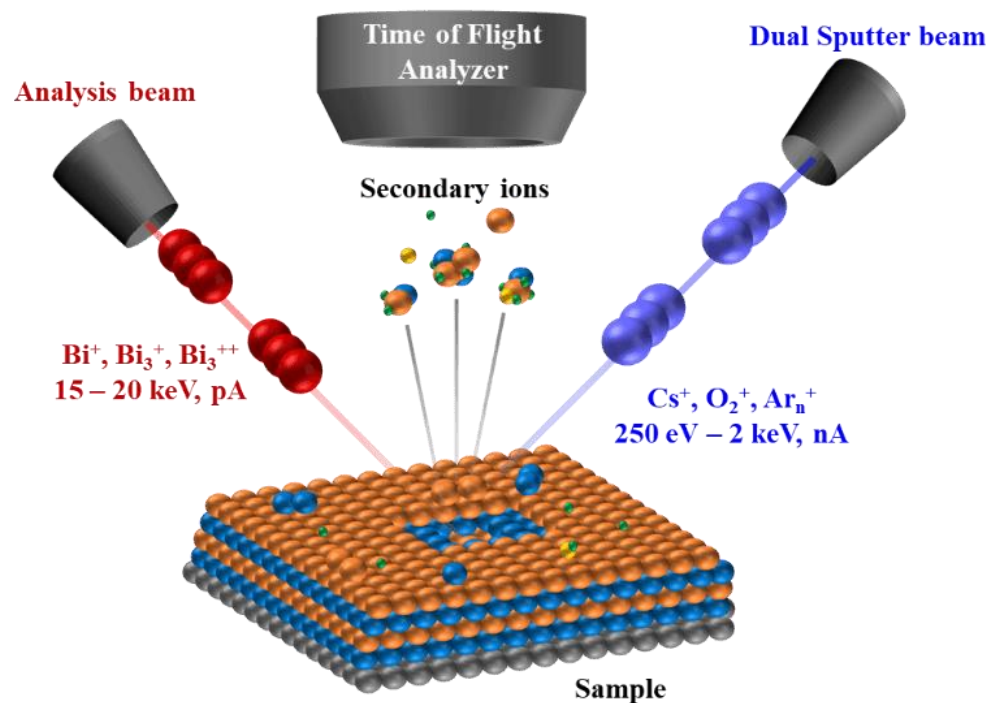
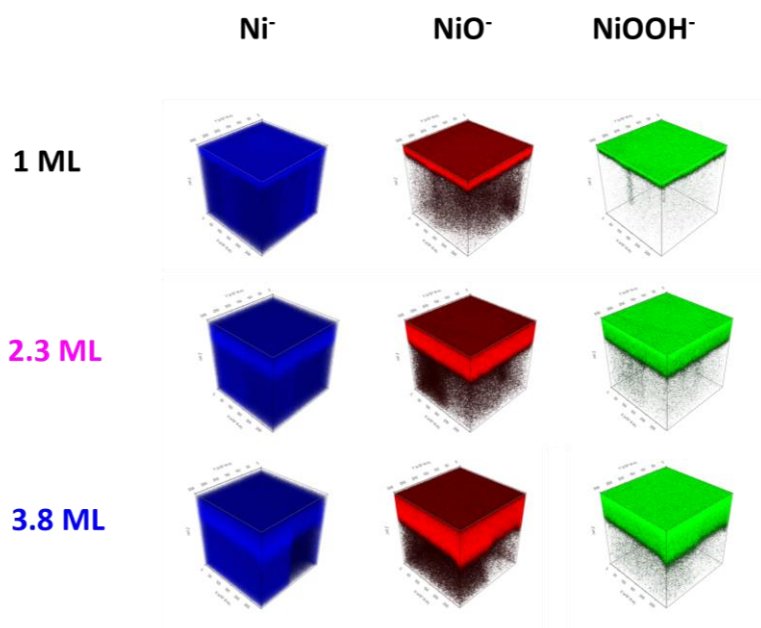


Electrocatalysis happens at the surface



- We are mainly interested in the first few layers of the electrode material.
- Voltammograms give us information about the interaction of electrode material with the electrolyte
- Polarization curves give us information about the catalytic activity of material for a given reaction, e.g. HER
- We want to correlate surface chemistry with electrochemical properties
- We need surface-sensitive analytical tools that provide us with surface composition of the electrode
- One of such tools is ToF-SIMS

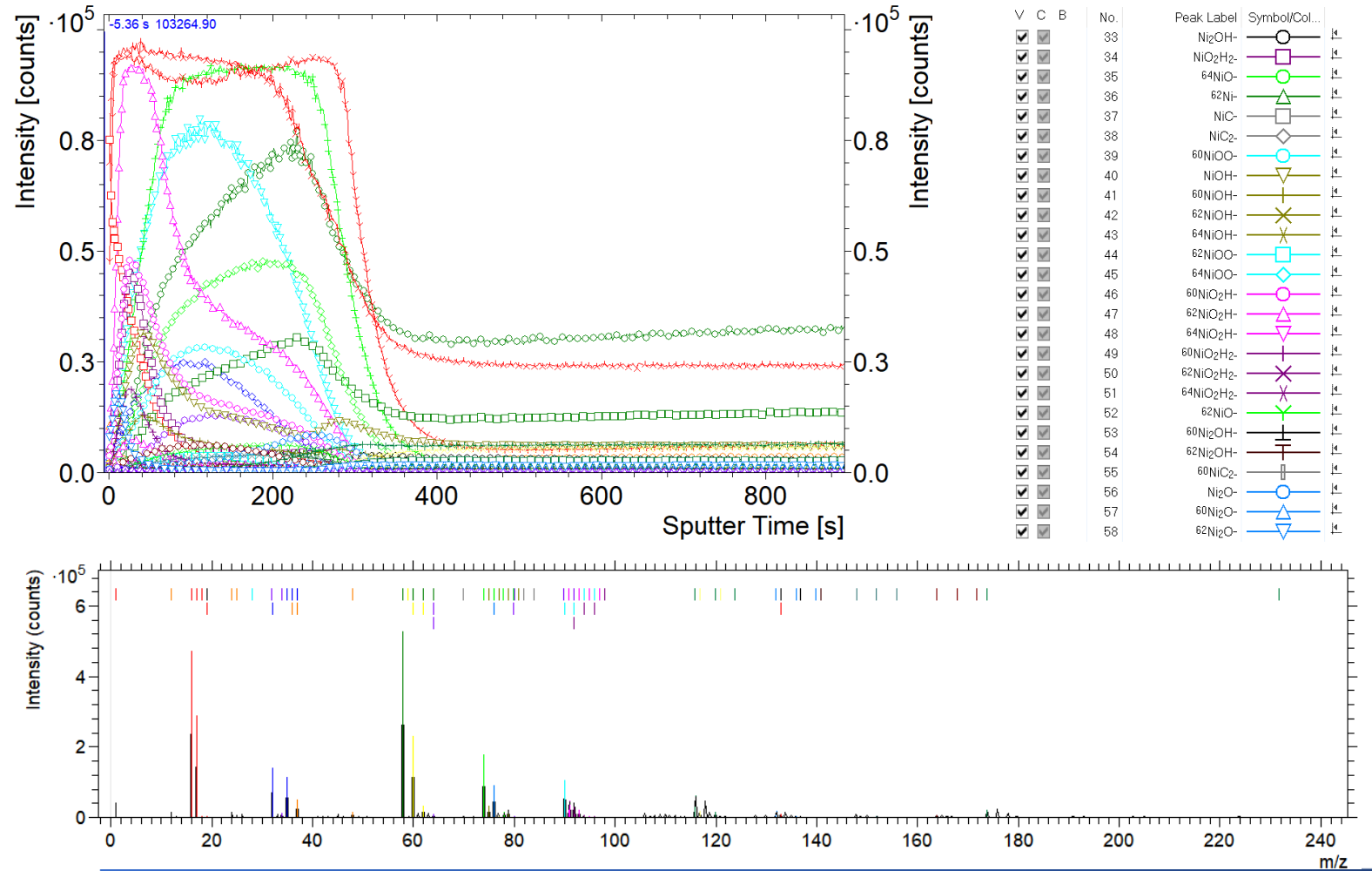
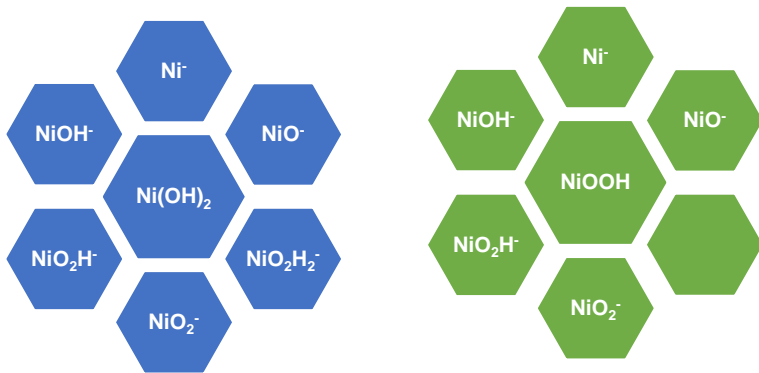
ToF-SIMS



- ToF-SIMS is one of the most sensitive tools available for surface analysis
- It provides information about chemical composition of the sample in 3D with nanometer depth resolution
- Ideal for needle in a haystack problems – we look for fragment with mass of analyte
- However,...

ToF-SIMS data

- We collect 500 spectra per depth profile
- In each spectra, there are over 1000 fragments identified
- Each fragment is formed from a species/compound in the sample
- However, same fragment can come from multiple sources and same source can create multiple fragments



Clustering based on correlation matrix

- We find three groups of fragments

C_3^- , C_2H^- , F^- , SO_3^- , C_4^- , C_3H^- , Cl^- , $^{37}Cl^-$, C_5^- , H^-
 $^{18}OH^-$, SO_2^- , NiO_2H^- , O^- , Cs^- , $^{60}NiO^-$, NiO^- , S^- , Si^- , AlO^- ,
 $NiOO^-$, $^{18}O^-$, O_2^- ,
 Zn^- , Ni^- , $^{60}Ni^-$, S_2^- , $^{60}NiH_2^-$, NiH_2^- , OH^- , C_2^- , NiH^-

- First group corresponds to surface contaminants during sample transfer
- Second group corresponds to the interfacial film grown electrochemically
- Third group corresponds to the bulk of the sample

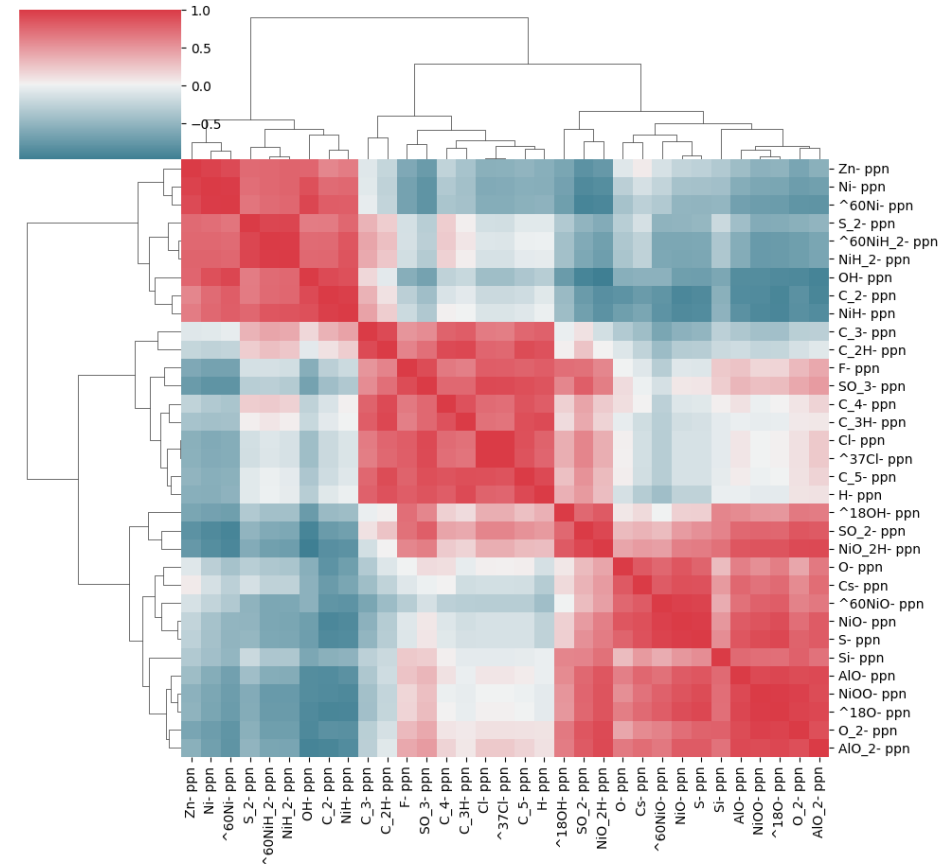


Fig. 1. The result of hierarchical clustering, which is based on a correlation matrix (i.e., distance determined by Spearman's coefficient of correlation) between the fragment concentrations. It highlights three prominent clusters (represented by red squares).

Equation discovery

- We use the symbolic regression method SINDy to identify mathematical relations between the fragments' intensities in mass spectra
- We work under the assumption that each fragment can be written as a linear combination of the other fragments
- We have some naturally built-in controls, i.e. isotope ratios, which the method identifies
- It also gives several nonsensical correlations and a few that need to be further investigated

RMSE	Equation
0.021795863084589678	$\text{NiO}_2\text{H} = 0.6 \text{ H} + 0.1 \text{ O}$
0.0033206752460522896	$\text{NiO}_2\text{H} = -1.1 \text{ }^{60}\text{Ni} + -1.0 \text{ H} + -0.2 \text{ NiOO} + -0.2 \text{ O} + 0.1 \text{ 1}$
0.004233586747772646	$\text{ }^{60}\text{NiO} = 0.4 \text{ NiO} + 0.3 \text{ NiOO}$
0.0037185370554185812	$\text{ }^{60}\text{NiO} = 1.7 \text{ S} + 0.4 \text{ NiO} + -0.2 \text{ NiOO} + -0.1 \text{ O} + 0.1 \text{ H}$
0.008058476834946827	$\text{NiO} = 1.4 \text{ }^{60}\text{NiO} + 0.2 \text{ O}$
0.007406643004447469	$\text{NiO} = 1.7 \text{ }^{60}\text{NiO} + -0.3 \text{ NiOO} + 0.2 \text{ O}$
0.015212412093664442	$\text{NiOO} = 0.8 \text{ }^{60}\text{NiO}$
0.0023918287909048186	$\text{NiOO} = 2.1 \text{ S} + 0.9 \text{ }^{60}\text{NiO} + -0.5 \text{ NiO} + 0.2 \text{ NiO}_2\text{H}$
0.001435557137991067	$\text{ }^{60}\text{Ni} = 0.4 \text{ Ni}$
0.003683555402379558	$\text{Ni} = 2.6 \text{ }^{60}\text{Ni}$
0.0030517186004265597	$\text{Ni} = 2.6 \text{ }^{60}\text{Ni} + -0.1 \text{ NiOO} + 0.1 \text{ NiO}_2\text{H}$
0.00017684283747765588	$\text{NiH}_2 = -0.3 \text{ S} + 0.2 \text{ C}_2\text{H} + 0.1 \text{ NiOO} + 0.1 \text{ H}$
5.814649192873677e-05	$\text{ }^{37}\text{Cl} = 0.3 \text{ Cl}$
4.4421552673896455e-05	$\text{C}_4 = 0.1 \text{ C}_2\text{H}$
0.01518639975828391	$\text{H} = 0.3 \text{ NiO}_2\text{H}$
0.0020718868588780735	$\text{O}_2 = 0.2 \text{ NiOO}$
0.03758364227543406	$\text{O} = 1.3 \text{ NiO}$
0.023018259177337828	$\text{O} = 1.4 \text{ NiO} + -1.1 \text{ NiOO} + 0.4 \text{ OH}$
0.0008732008268265299	$\text{S} = 0.2 \text{ NiOO} + 0.1 \text{ NiO}$
0.051956297621226194	$\text{OH} = 0.6 \text{ O}$
0.0045469638968045935	$\text{OH} = 0.6 \text{ NiOO} + -0.4 \text{ NiO} + 0.1 \text{ 1} + 0.1 \text{ Ni} + 0.1 \text{ O}$

Conclusions

- Initial application of machine learning methods on ToF-SIMS data has given some encouraging preliminary results
- Clustering of fragments shows logical distribution of species within the electrochemical interface, significantly reducing time of similar analysis if done by user
- Equation discovery needs further improvement, most likely better initial guidance, but we also need some improvements in spectra collection and sample preparation
- **Quality of data is the most important factor in successful implementation of ML methods**



Thank you!

